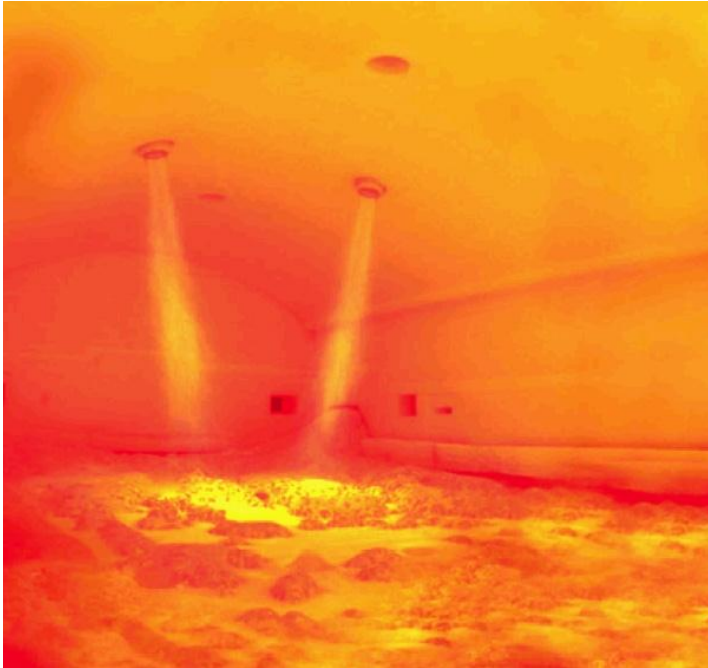


glass melting process => 80% of total energy

=> 50% of total cost



The Comparison of Energy Consumption between the Soda-lime Glass and Borosilicate Glass Furnace

KANIT TAPASA

**Thailand Center of Excellence for
Glass**

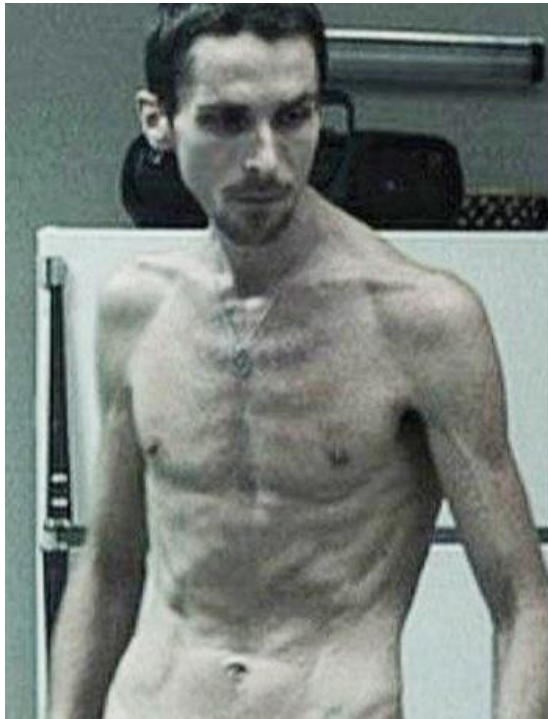
**Department of Science Service,
Thailand**



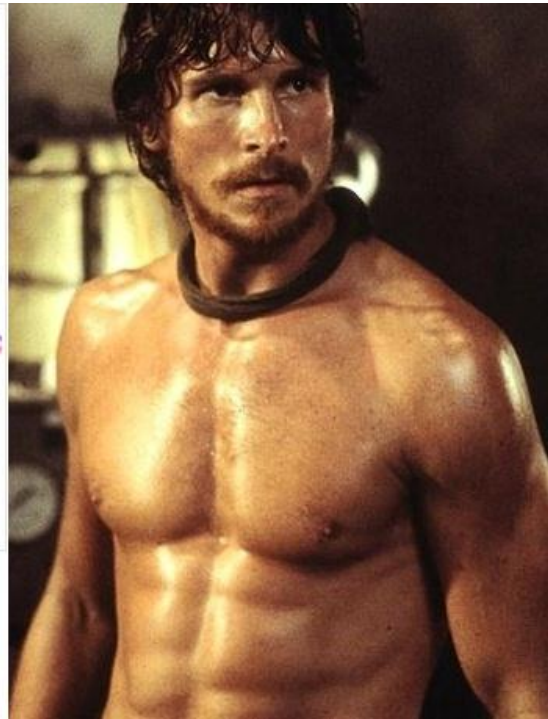
- Mission: To provide scientific and technical supports to the glass industry in Thailand
- Project: Energy reduction glass melting furnace
 - Batch modification
 - Furnace performance assessment

Energy reduction plan

BEFORE



AFTER

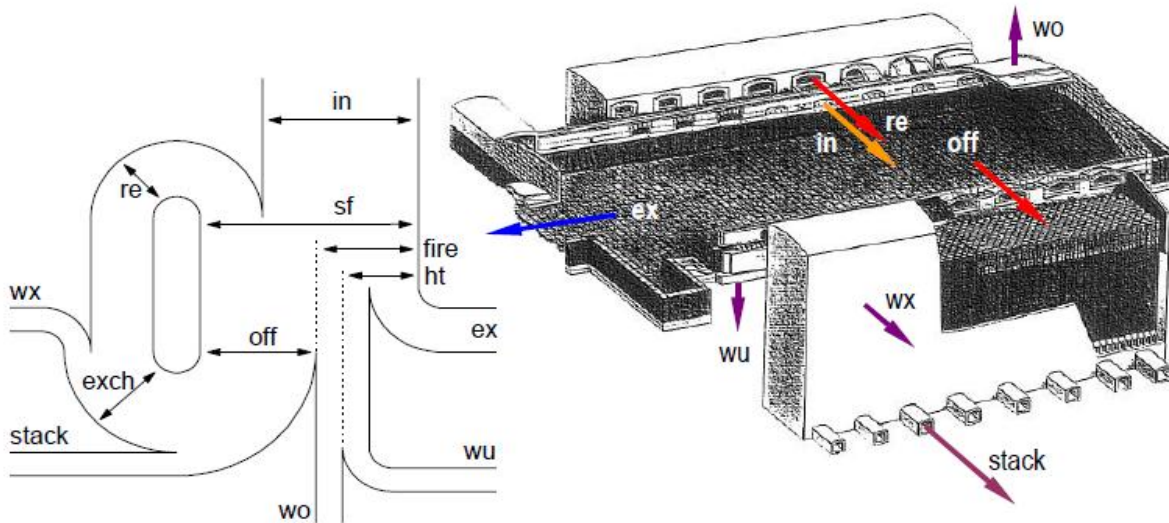




Glass furnace performance



Heat balance

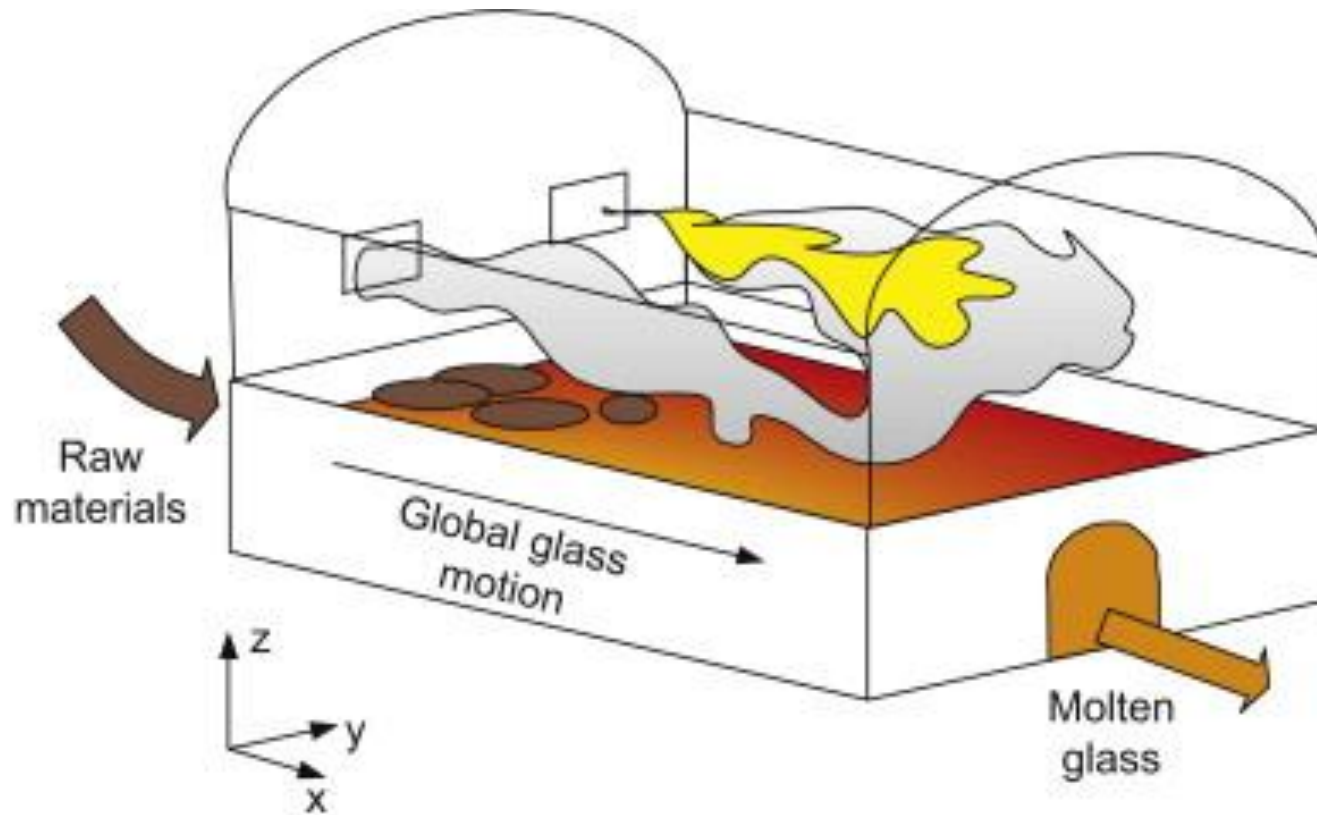


Outline

- Calculations
- Results
- Conclusions

Calculations

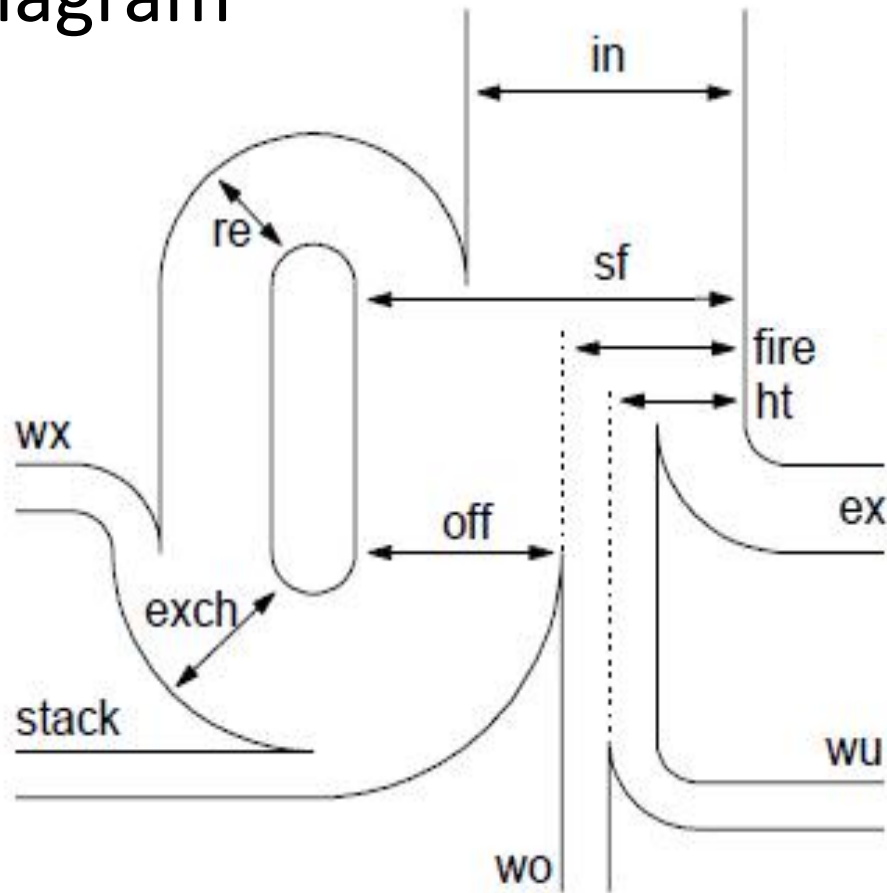
End-port furnace



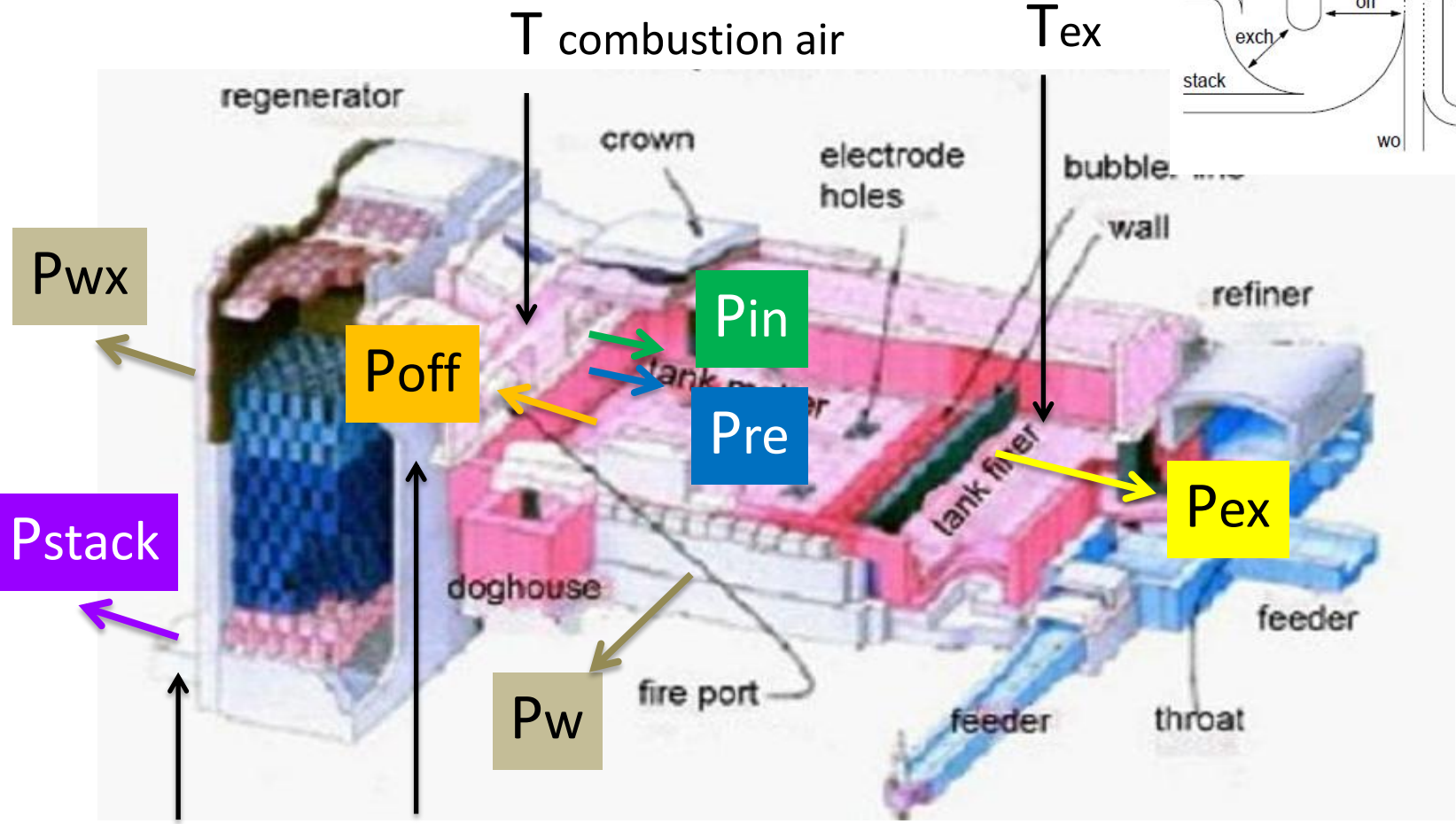
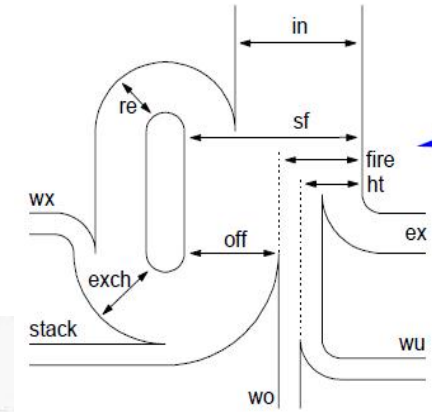
Source: [Olivier Achet et al. Control Engineering Practice Volume 16, Issue 12, 2008.](#)

Heat balance

- Sanky diagram



Heat balance



$$P_w = P_{in} + P_{re} - P_{off} - P_{ex}$$

$$P_{wx} = P_{off} - P_{stack} - P_{re}$$

Exploited heat (Hex)

$$H_{ex} = (1-y_c) \Delta H^{\circ}_{chem} + \Delta H_{melt}(T_{ex})$$

ΔH°_{chem} is the heat required for the batch-to-melt conversion at 298 K.

$$\Delta H^{\circ}_{chem} = H^{\circ}_{glass} + H^{\circ}_{gas} - H^{\circ}_{batch}$$

batch (298 K) \rightarrow glass (298 K) + batch gases (298 K)

$\Delta H_{melt}(T_{ex})$ is the heat content in melt at T_{ex} .

$$\begin{aligned} \Delta H_{melt}(T_{ex}) &= H_{T,melt} - H^{\circ}_{glass} \\ &= H^{\circ}_{1400,melt} + c_{p,melt} \cdot (T-1400) \end{aligned}$$

Batch formula

Data from a factory

y_c is % cullet

$$H_{ex} = (1 - y_c) \Delta H^{\circ}_{chem} + H^{\circ}_{1400,melt} + C_{p,melt} (T_{ex} - 1400)$$

$$P_{ex} = H_{ex} \times \text{pull}$$

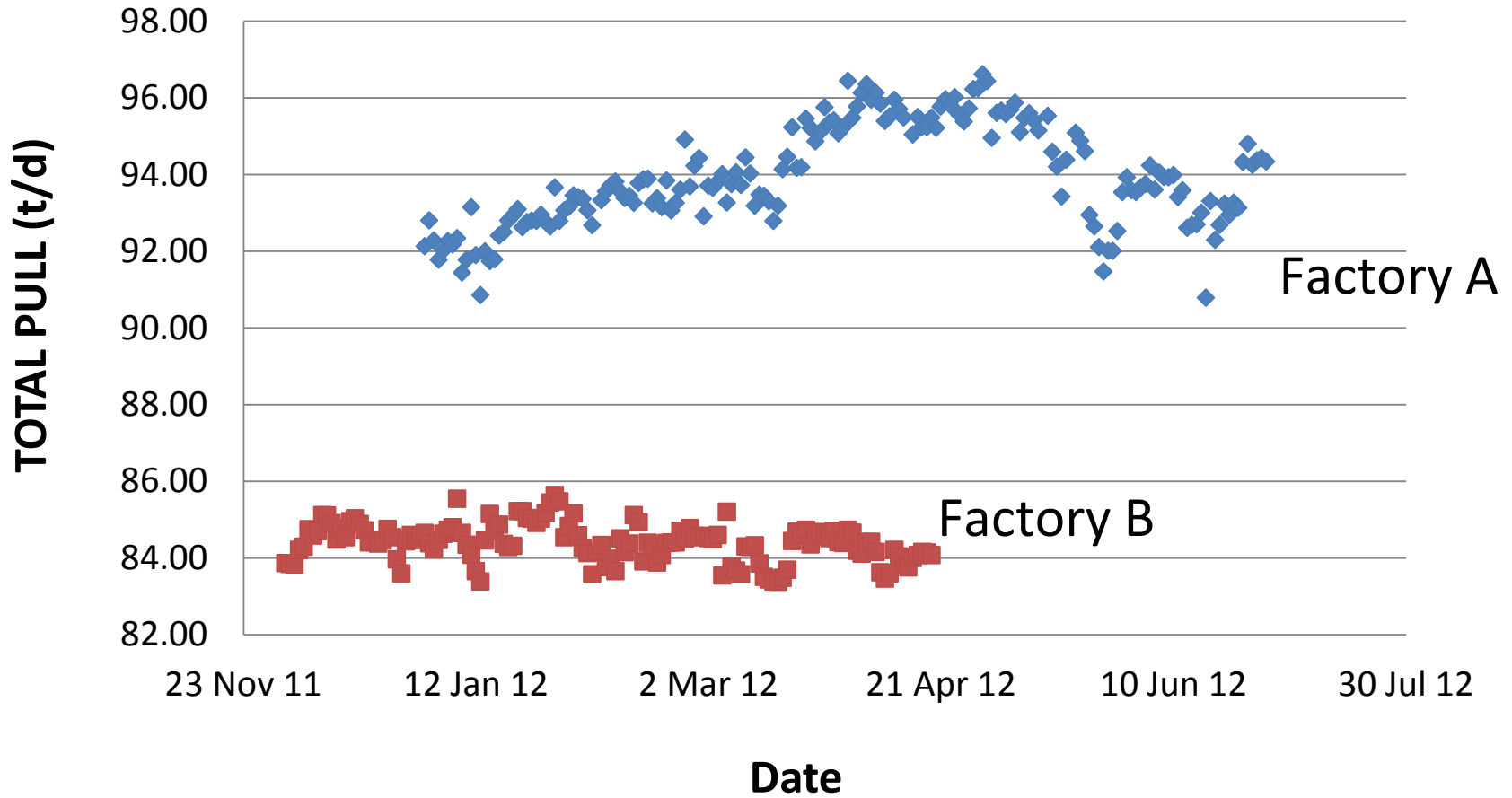
Data

DATE	TOTAL PULL (t/d)	MELT TEMP. (°C) Tex	FLUE GAS TEMP. (°C)		COMBUSTION AIR TEMP. (°C) Tre	AIR/GAS RATIO	FUEL m ³ /h	SECONDARY AIR m ³ /h	ELEC. (If use) kWh/h	OXYGEN (If use) m ³ /h	% CULLET
			Toff	Tstack							
1 Jun 11	104.74	1342	1321	407	1289	10.60	668	6757	216		20
2 Jun 11	105.52	1333	1322	409	1291	10.60	675	6869	214		20
3 Jun 11	105.74	1340	1333	410	1298	10.60	678	6837	214		20
4 Jun 11	105.65	1340	1330	409	1296	10.61	674	6812	214		20
5 Jun 11	105.65	1342	1328	406	1296	10.60	672	6804	214		20
6 Jun 11	105.74	1338	1328	406	1298	10.60	676	6846	214		20
7 Jun 11	105.00	1341	1330	407	1296	10.60	677	6833	216		20
8 Jun 11	105.62	1339	1324	409	1291	10.61	671	6792	213		20
9 Jun 11	105.91	1339	1321	410	1288	10.60	675	6846	214		20
10 Jun 11	105.48	1336	1322	410	1289	10.60	679	6886	213		20
11 Jun 11	105.54	1344	1326	408	1292	10.60	681	6888	216		20
12 Jun 11	106.29	1344	1328	409	1292	10.60	682	6897	212		20
13 Jun 11	105.37	1346	1339	405	1304	10.60	681	6871	220		20
14 Jun 11	104.25	1346	1332	406	1296	10.59	673	6802	218		20
15 Jun 11	103.88	1344	1327	404	1290	10.60	665	6679	215		20
16 Jun 11	103.12	1341	1321	401	1285	10.60	658	6652	216		20
17 Jun 11	103.19	1340	1322	399	1283	10.61	658	6653	215		20

Furnace data

	Factory A	Factory B
Furnace Type	Regenerative end-port furnace	Regenerative end-port furnace
Area	52 m ²	60 m ²
Capacity	120 ton	200 ton
Total Pull	90.8 – 96.6 (ton/day) 3.78 – 4.03 (ton/hour)	83.4 – 85.6 (ton/day) 3.48 – 3.57 (ton/hour)
Input energy	NG + Electricity	NG
Date of data collection	1 Jan - 30 June 12 (6m)	1 Dec 11 – 19 April 12 (5m)

Total pull (t/d)



Batch data

Raw materials	H° (kWh/kg)*	Amount of Raw Materials per 1 ton of glass (kg)	
		Factory A	Factory B
Silica sand (SiO ₂)	4.2112	685	628
Sodium Feldspar (NaAlSi ₃ O ₈)	4.1649	51	110
Potassium feldspar (KAlSi ₃ O ₈)	3.9635	12	-
Dolomite (CaMg(CO ₃) ₂)	3.4873	193	161
Sodium Carbonate (Na ₂ CO ₃)	2.9608	231	250
Borax .5 H ₂ O (Na ₂ O·2B ₂ O ₃ ·5H ₂ O)	4.5676	-	14
Potassium Carbonate (K ₂ CO ₃)	2.3117	-	13

Glass components

components	%wt	
	Factory A	Factory B
SiO ₂	72.45	67.94
TiO ₂	0.03	0.06
Al ₂ O ₃	1.38	2.10
B ₂ O ₃	-	0.54
Fe ₂ O ₃	0.11	0.09
P ₂ O ₅	0.01	0.01
MgO	3.38	3.67
CaO	7.27	6.53
Na ₂ O	15.06	17.54
K ₂ O	0.30	1.28

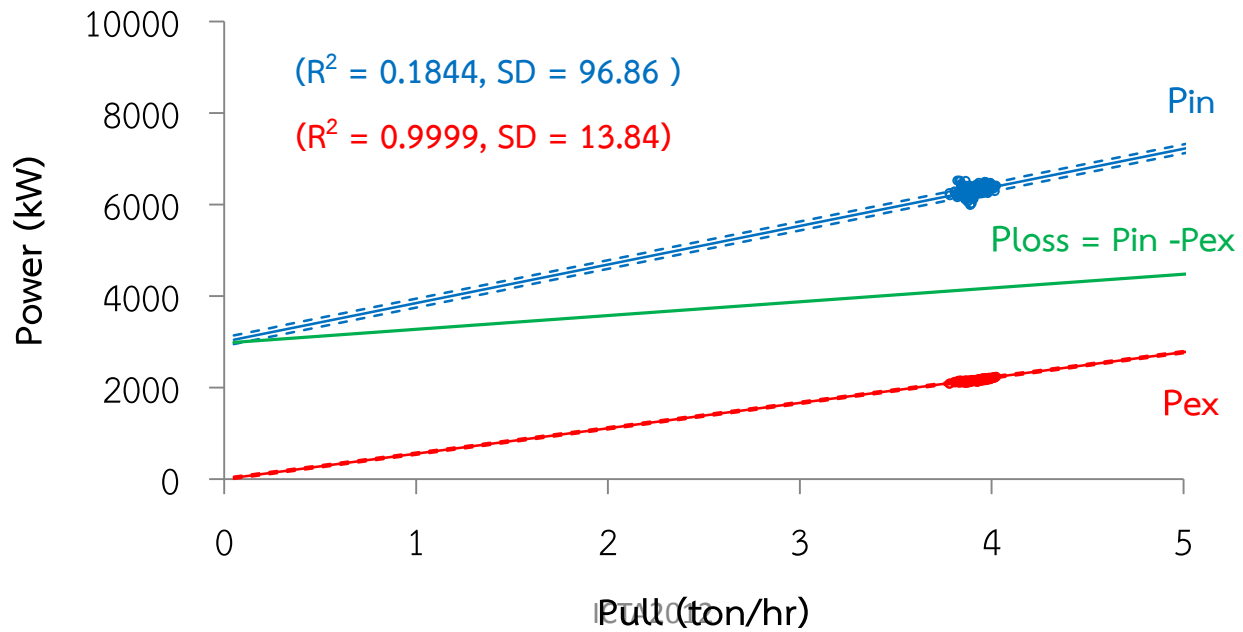
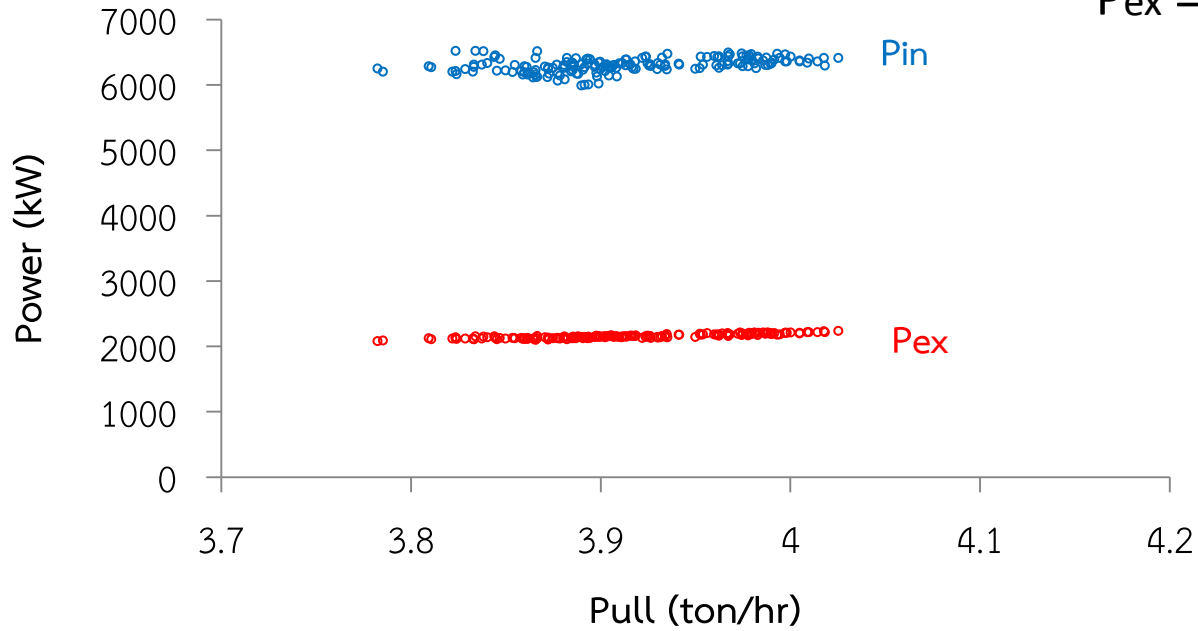
Batch data

<i>kWh/t</i>	Factory A	Factory B
H°_{batch}	4517.69	4366.34
H°_{glass}	3930.81	3865.61
H°_{gas}	437.34	362.71
$\Delta H^{\circ}_{\text{chem}}$	149.54	138.02

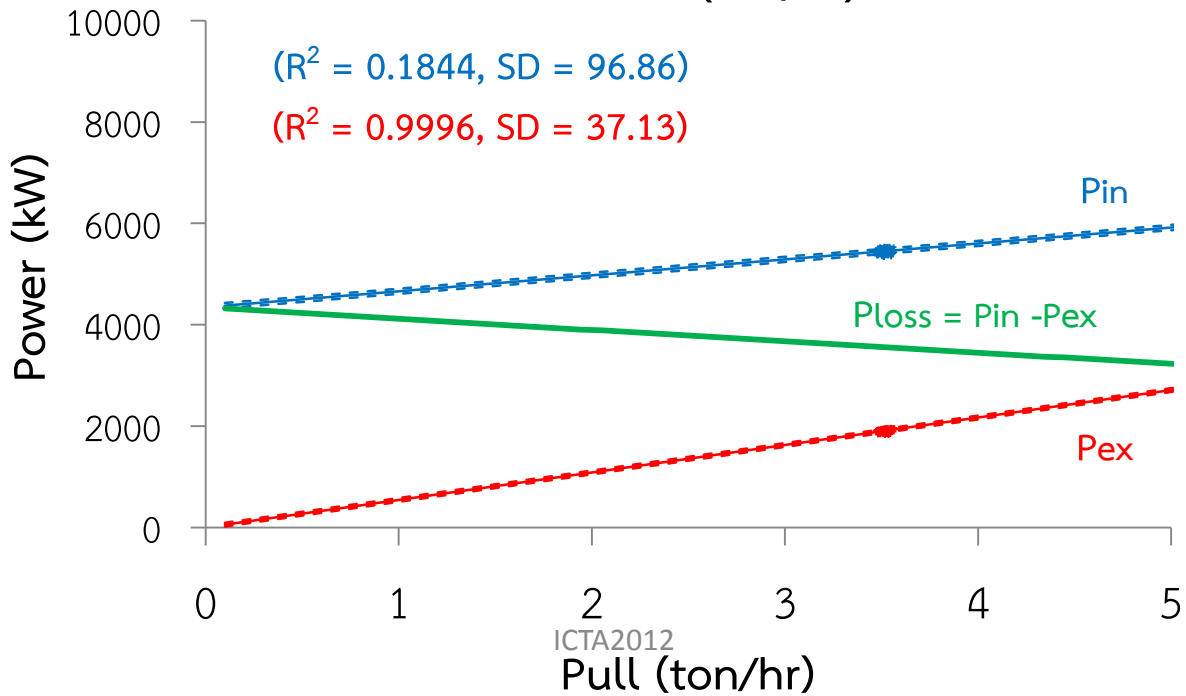
Results

Factory A

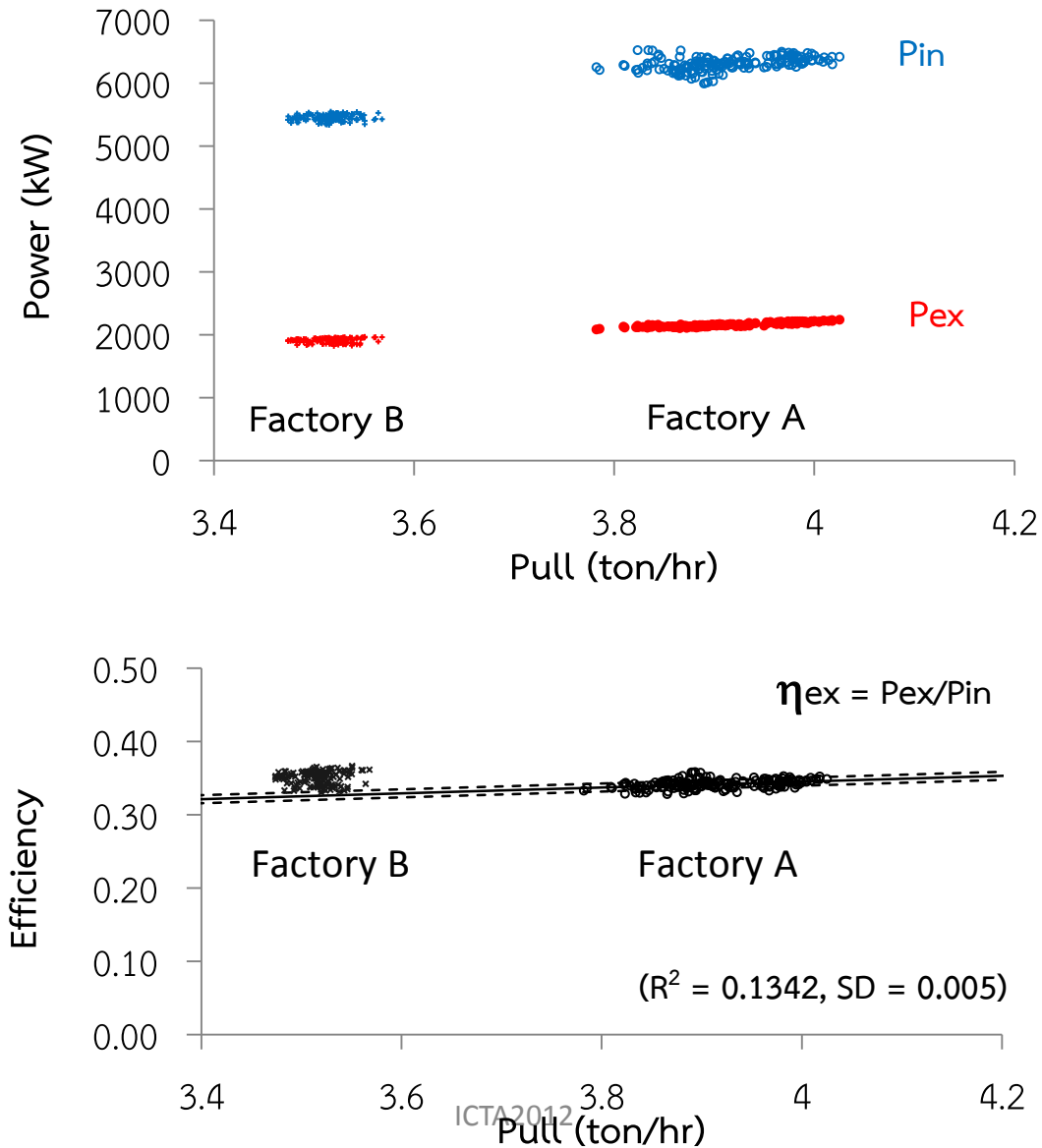
$$P_{ex} = H_{ex} \times \text{pull}$$

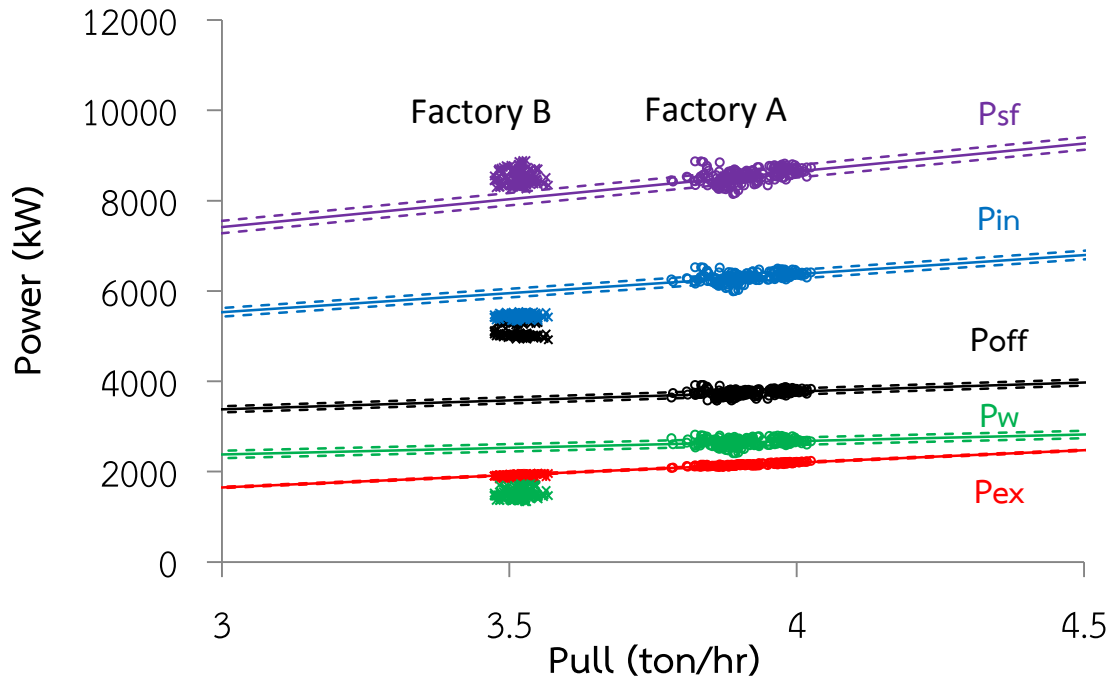


Factory B



Furnace A and B





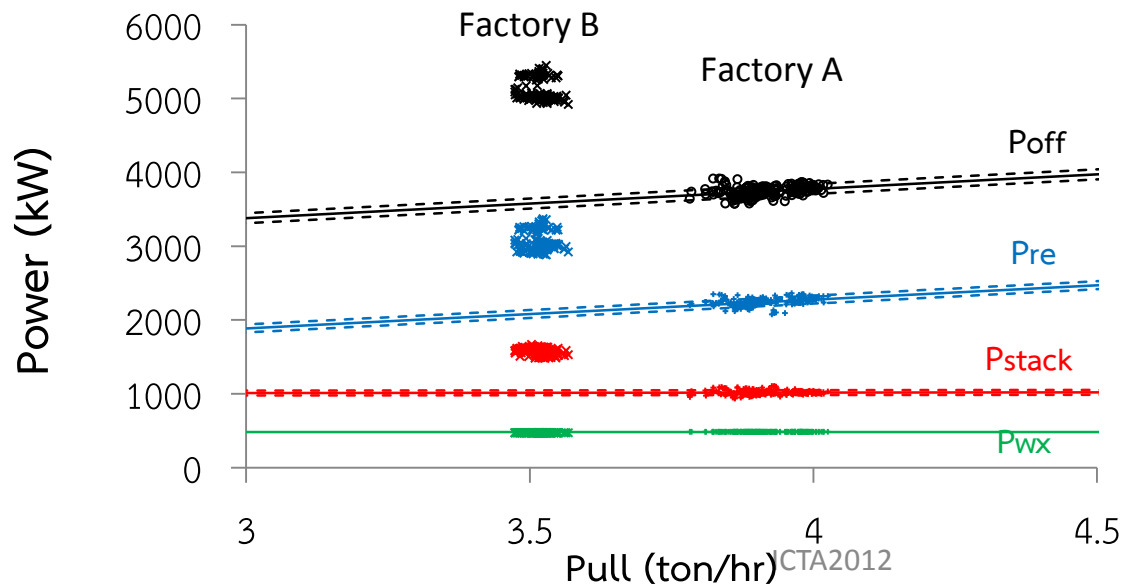
($R^2 = 0.1342$, $SD = 0.005$)

($R^2 = 0.1844$, $SD = 96.86$)

($R^2 = 0.0911$, $SD = 68.30$)

($R^2 = 0.0367$, $SD = 82.56$)

($R^2 = 0.9999$, $SD = 13.84$)



($R^2 = 0.0911$, $SD = 68.30$)

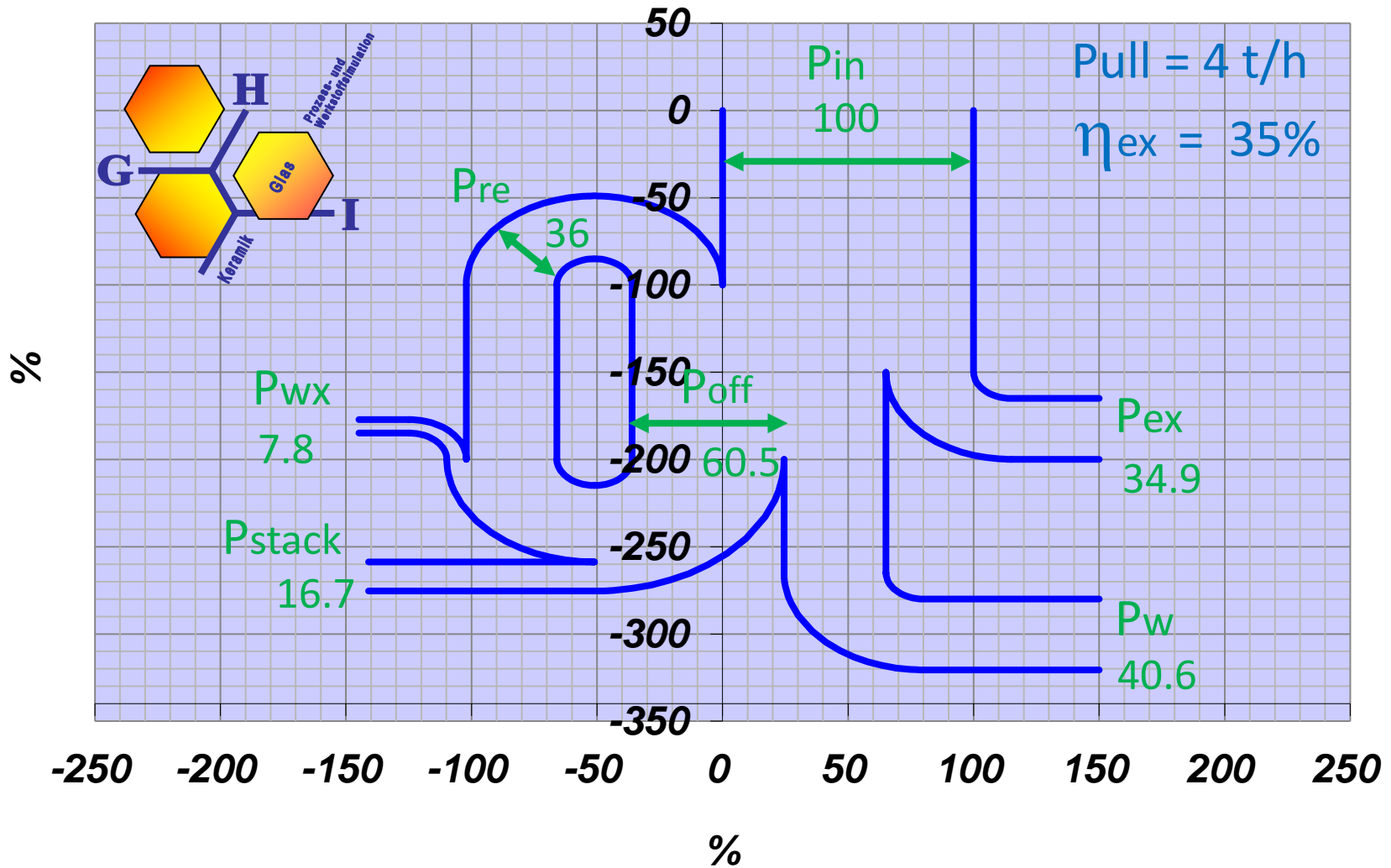
($R^2 = 0.1332$, $SD = 54.32$)

($R^2 = 0.0001$, $SD = 33.26$)

($R^2 = 1$, $SD = 0$)

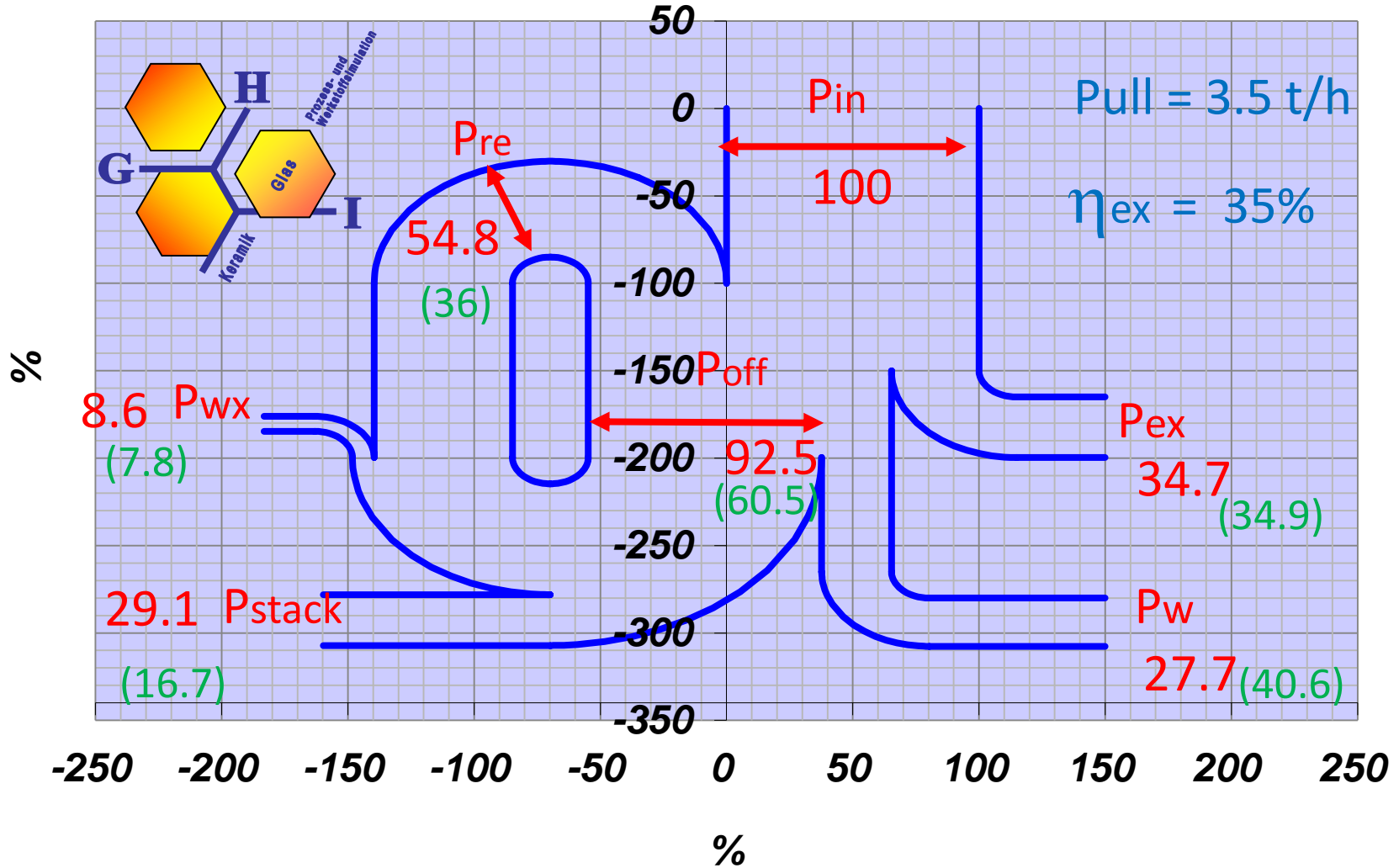
$P_{re} = 0.6P_{off}$

Factory A



$$P_{pre} = 0.6P_{off}$$

Factory B



$Pre = 0.6P_{off}$

Conclusions

- Heat balance (furnace performance) can indicate the situation of glass furnace in term of energy consumption.
- Furnace A has 40 - 43% wall losses and 60% heat loss from offgas.
- Furnace B has 25 - 31% wall losses and 92-98% heat loss from offgas.

Acknowledgement

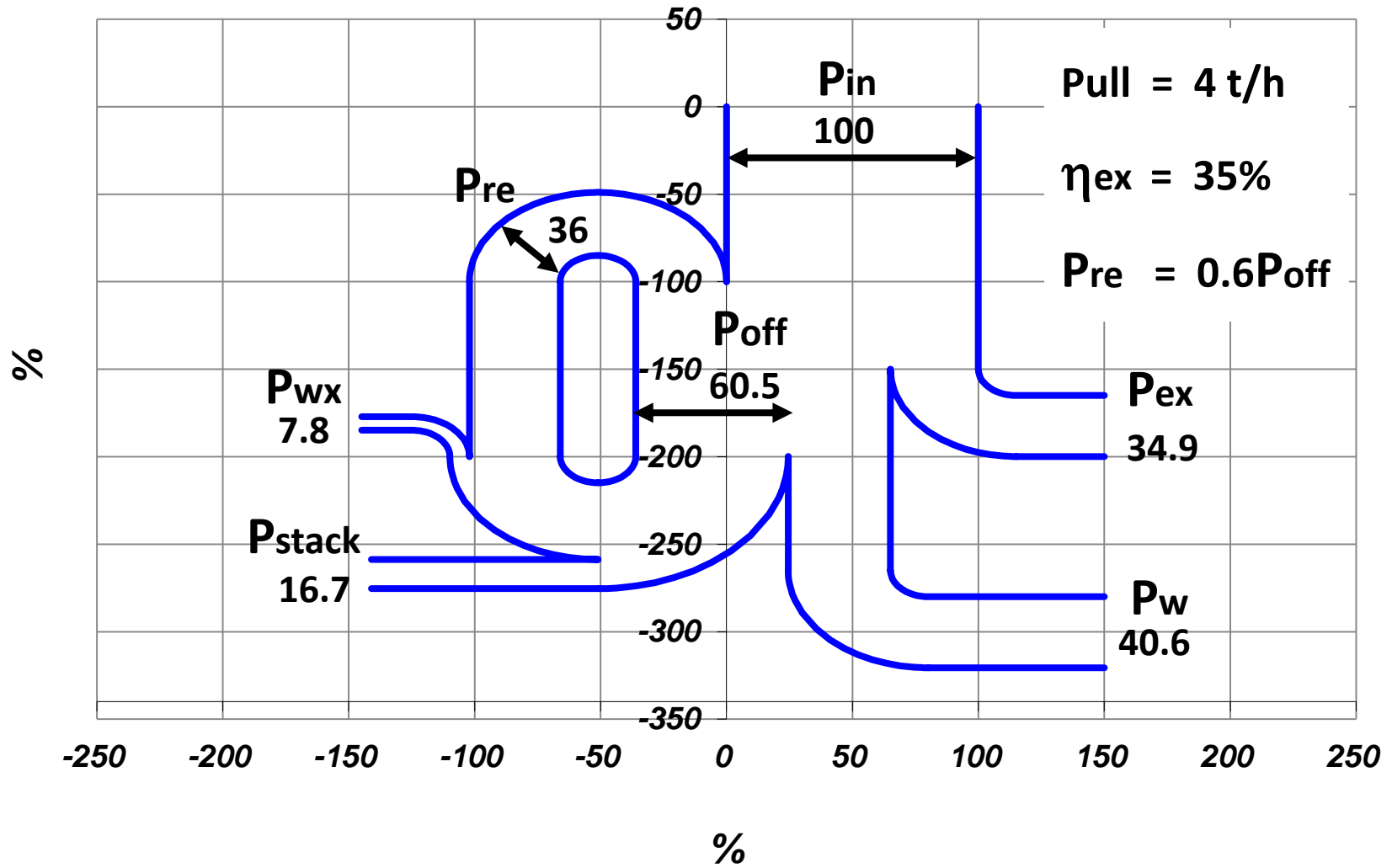
Prof. Reinhard Conradt

RWTH Aachen University, Germany



Thank you for your attention

Furnace A



Furnace B

