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ASSOCIATION OF FOOD SCIENTISTS & TECHNOLOGISTS, INDIA

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ASSOCIATION OF FOOD SCIENTISTS AND TECHNOLOGISTS

(INDIA)

The Association is a professional and educational organization of Food Scientists and Technologists

AFFILIATED TO THE INSTITUTE OF FOOD TECHNOLOGISTS, USA

Objects:

- 1. To stimulate research on various aspects of Food Science and Technology.
- 2. To provide a forum for the exchange, discussion and dissemination of current developments in the field of Food Science and Technology.
- 3. To promote the profession of Food Science and Technology.

The ultimate object is to serve humanity through better food.

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- 2. Arranging lectures and seminars for the benefit of members.
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RESEARCH PAPERS

Nutritive Value of Soy Idli*

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Manuscript received 28 October 1981; revised 30 April 1982

Nutritive value of soy idli was studied using rat growth experiment. Weaning albino rats were fed diet consisting wholely of either unfermented, or fermented soy idli supplemented with lime powder and fenugreek leaves. The control group was fed 20% casein diet. Fermented soy idli was found superior to unfermented soy idli in terms of body weight gain, hepatic protein and vitamin content and femur composition. Supplementation of fermented soy idli with lime and fenugreek leaves further improved the growth, liver and femur composition and was comparable with the control group (20% casein diet) thereby indicating that fermented soy idli along with lime and greens can form a good supplement for undernourished children.

It is well known that legumes form a valuable supplement to cereal diet especially when the diet is lacking in animal proteins¹⁻⁴. Although soybean is superior with regard to nutrient content compared to legumes used in our country, it is not used in any appreciable amounts. The special processing needed to remove the toxic factor and beany flavour presents a problem. In Far East, fermented foods prepared from soybean are popular⁵⁻⁷. These foods are mainly fermented by molds which impart a chracteristic flavour not quite familiar to the Indian palate. On the other hand, fermented foods of India such as idli *khaman* and *dhokla* are popular and involve only short and simple process of fermentation.

Previous studies have shown that fermentation brings about changes in the food which improves taste, texture⁴⁻⁷, nutrient content such as **B** vitamins⁸⁻¹⁴ and also nutritive value and digestibility⁶8,¹⁵⁻¹⁹.

Replacement of traditional ingredients with less popular cereals/millets and legumes has been successfully achieved in this laboratory^{4,20}. We have been successful in substituting soybean for the traditional black gram in the preparation of idli with improved chemical composition²¹. Present studies were therefore, carried out to determine the nutr tive value of fermented scy idli as compared to unfermented soy idli by rat growth experiment. Incorporation of leafy vegetable and lime powder (a mixture of CaCO₃, CaO and Ca (OH)₂ providing 60% calcium, commonly known as *chuna*) in acid foods such as fermented foods has been shown to give good availability of calcium and vitamin $A^{20,22}$. The effect of such incorporation on nutritive value of soy idli was also studied.

Materials and Methods

The experimental diets were prepared as follows: 1000 g of coarsely ground rice flour, 500 g of finely ground soydal flour and 50 g of common salt were mixed thoroughly with 2500 ml of water. Idlis were prepared by steaming for 5 min in an idli steamer both with unfermented batter and batter fermented at 30°C for 16 hr. One batch of fermented batter was supplemented with 330 g of chopped fenugreek leaves and 5.0 g of lime powder (commercially available chuna powder used with betel leaves which is a mixture of CaCO₃, CaO and Ca $(OH)_2$). Addition of lime powder did not raise the pH beyond 6.5 and contributed 225 mg calcium/100 g of soy idli mix. The samples were freeze dried. These preparations formed the entire diet without any supplementation. The control diet of 20 per cent casein was prepared as described earlier²³.

Weaning albino rats of Charles Foster strain were used in these studies. Equal number of animals of both the sexes were used in each group. The rats were divided in to six groups and fed following diets *ad libitum* for 4 weeks. (I) 20 per cent casein diet, (II) unfermented soy idli, (III) fermented soy idli, (IV) fermented soy idli supplemented with fenugreek leaves and lime, (V) diet III fed along with group II (Pair-fed) and (VI) diet IV fed along with group II (Pair-fed). Daily food intake and weekly gain in body weight were recorded.

At the end of four week feeding, the animals were

^{*} Part of these studies was presented at 47th Annual Meeting of Society of Biological Chemists (India) at Delhi, October, 1978. Present Address: Biological Chemistry Division, Cancer Research Institute, Tata Memorial Centre, Parel, Bombay-400 012.

killed by decapitation under light with ether anesthesia. The abdomen was opened, the liver was removed quickly, freed of adhering blood and connective tissues, weighed and taken for the estimation of protein, thiamin, riboflavin and vitamin A.

Liver protein was determined by the method of Lowry $et al.^{24}$ and thiamin, riboflavin and niacin by the methods described by the Association of Vitamin Chemists²⁵ Vitamin A was estimated by the modified method of Gallup and Hoefer²⁶.

The femurs were removed from both the sides and freed of adhering connective tissues and muscles and used for the analysis of moisture, fat, ash and calcium content by A.O.A.C. methods²⁷.

Results and Discussion

The data on food intake and body weight gain are given in Table 1. The group fed fermented soy idli showed increased food intake over the group fed unfermented soy idli indicating increased acceptability of the former. The food intake was further increased by incorporation of fenugreek leaves and lime. The rats fed fermented soy idli showed higher body weight gain than those fed unfermented soy idli. The increase in body weight was further improved by incorporation of leafy vegetable and lime. Weight gain per gram protein was higher in rats fed fermented soy idli as compared to that of rats fed unfermented soy idli. Even the pair-fed group showed higher weight gain per gram

TABLE 1. FOOD INTAKE AND BODY WEIGHT GAIN IN RATS FED SOY IDLI DIET										
	20% casein	Unfermented soy idli	Fermented soy idli	Fermented soy idli + greens + lime	diet III+II	diet IV+111				
	(I)	(II)	(III)	(IV)	(V)	(VI)				
No. of animals	9	9	8	9	9	9				
Protein in diet (g)	20.0	19.5	19.5	19.9	19.5	19.9				
Dietary intake/4 Wk	c .									
Food (g)	222.0 ± 4.0	117.0 ± 2.0	177.0 ± 2.0ª	222.0 ± 3.0	117.0 ± 2.0	117.0 ± 2.0				
Energy (Cal)	912.0 ±15.0	426.0 ± 7.0	610.0 ± 6.0^{a}	755.0 <u>+</u> 9.0	403.0 ± 6.0	399 ± 10				
Protein (g)	44.0 ± 1.0	23.0 \pm 0.4	34.0 ± 0.3^{a}	44.0 ± 0.5	23.0 ± 0.4	$23.0~\pm~0.3$				
Body Wt. gain* (g)									
for 4 Wk	73.0 ± 3.5	19.0 ± 0.9	40.0 ± 1.7^{a}	70.0 ± 2.0	26.0 <u>+</u> 0.7ª	33.0 ± 0.8 ^b				
per 100 Cal.	8.1 ± 0.28	4.6 ± 0.18	6.6 ± 0.23^{a}	9.2 ± 0.16	6.2 ± 0.16^{a}	8.5 ± 0.21				
per g protein	1.67± 0.06	0.87 ± 0.03	1.19± 0.04ª	1.60 ± 0.03	1.11 ± 0.03ª	1.47 ± 0.40				

Values are mean \pm S.E.

*Initial body weight ranged between 45-50 g.

^aValues significantly different from group II values (p. <0.002)

^bValues significantly different from group V values (p < 0.002)

		TABLE 2. LIVER CO	MPOSITION OF RA	TS FED SOY IDLI DIET		
	20% casein	Unfermented sov idli	Fermented soy idli	Fermented soy idli +	Diet III+II	Diet 1V+11
	(I)	(II)	(III)	(IV)	(V)	(VI)
Final Body wt. (g)	120.00±3.6	66.00±1.5	87.00±1.6ª	117.00 ± 2.3	73.00±1.3ª	81.00±1.5
Liver wt. (g)	4.30±0.20	2.10±0.29	3.50±0.12ª	4.60 ± 0.22	2.69±0.12ª	3.00±0.08
Protein (g)*	20.80 ± 0.47	14.30±0.47	18.30±0.43ª	20.10±0.33	16.40 ± 0.25^{a}	17.90±0.21b
Vitamin A (mg)*	10.10 ±0.31	4.90 <u>+</u> 0.16	6.20±0.23ª	10.60±0.39	5.10±0.21	6.80±0.29b
Thiamin (mg)*	1.10±0.055	0.49±0.026	1.05±0.064ª	1.11±0.088	0.69±0.028ª	0.79 <u>+</u> 0.044
Riboflavin (mg)*	3.39 <u>+</u> 0.12	1.62±0.14	3.11±0.17ª	3.34±0.26	2.49±0.087ª	2.51 ±0.083
Niacin (mg)*	2.57 <u>+</u> 0.12	1.39±0.063	2.85 ± 0.05 ^a	3.11 ±0.40	2.04 ± 0.067^{a}	2.29+0.073

*Values are for 100g liver; values are Mean \pm S.E.

^aValues significantly different from group II values (p<0.002)

^bValues significantly different from group V values (p < 0.001)

	20% casein	Unfermented soy idli	Fermented soy idli	Fermented soy idli + greens + lime	Diet III+II	Diet IV+III
	(1)	(II)	(III)	(IV)	(V)	(VI)
Wet wt. (mg)	357.0 <u>±</u> 6.2	233.0±6.9	278.0±6.6	337.0 <u>+</u> 9.5	236.0 ± 3.3	278.0±4.7°
Dry wt. (mg)	191.0±2.4	114.0 ± 2.1	145.0±2.0	176.0±2.8	129.0±2.2ª	141.0±1.1°
Fat free dry wt. (mg)	182.0 <u>+</u> 1.9	104.0 <u>±</u> 2.9	137.0±1.7	167.0 <u>+</u> 2.6	121.0±2.2ª	133.0±1.0°
Ash wt. (mg)	88.0±4.0	48.0 <u>±</u> 4.1	71.0±1.1	91.0±2.6	59.0±2.5ª	70.0±1.2 ^c
Calcium (mg)	43.4 <u>+</u> 0.78	17.5 <u>±</u> 0.60	26.7 <u>+</u> 0.37	41.6 <u>+</u> 0.85	22.0±0.56 ^b	27.6 ± 0.84^{d}
Calcium (mg 100 g						
body wt.)	36.2±1.22	26.4 ±0.67	30.1±0.48	35.4 <u>+</u> 1.14	30.5±1.08 ^b	33.9 ± 1.27^{d}
Values are mean \pm	S.E.					
Values significantly	different from	group II values, a is	s p<0.001 and b	is p<0.01		
Values significantly	different from	group V values, c is	p<0.002 and d	is p<0.05		

TABLE 3. FEMUR COMPOSITION OF RATS FED SOY IDLI DIET

protein than those fed unfermented soyidli. This shows that fermentation improves the nutritive value of sov idli. Similar observations on fermented foods have been made by other investigators^{8,16-19}. Supplementation with greens and lime further improves the nutritive value of soy idli. The weight gain per gram protein of soy idli supplemented with greens and lime was comparable with 20 per cent casein diet. The supplementary effects of proteins from greens have been shown by Phansalkar et al²⁸. Apart from this, the greens also provide other nutrients such as carotene and iron⁴. Children given a supplement of *dhokla* containing greens and lime powder have shown to gain more weight than those receiving *dhokla* without any supplement²⁰.

Liver weights of rats fed fermented soy idli were higher than those fed unfermented soy idli (Table 2). Liver weight of the rats fed fermented soy idli with greens and lime showed further improvement and was comparable with that of rats fed 20 per cent casein diet. Liver status of the animals as judged by protein, vitamin A, thiamin, riboflavin and niacin contents was better in case of animals fed fermented soy idli. Liver vitamin A of the animals fed fermented soy idli with greens and lime was higher than those receiving diets without such supplement. The availability of vitamin A from greens have been shown earlier⁴. Improved liver status of the rats fed fermented idli and khaman has been shown in previous studies in this laboratory⁸. The biochemical status of the pre-school children given supplement of dhokla and dhokla with greens and lime was much superior to that of controls²⁰.

Composition of the femur of the rats fed soy idli is given in Table 3. The rats fed fermented soy idli showed better bone status than those fed unfermented soy idli. Incorporation of lime further improves the femur weight, ash and calcium content. This indicates that various nutrients such as calcium, phosphorus and vitamins have good availability in diet supplemented with lime and greens. Incorporation of lime in foods with acidic pH have been achieved successfully in earlier studies in this laboratory²³.

Earlier studies have shown that the fermentation improves the physico-chemical characristics of soy idli²¹. Present studies have further shown that fermentation also imrpoves the biological value as well as the availability of some of the vitamins. Fermented soy idli supplemented with greens and lime can be used as cheap supplement to pre-school children through feeding programmes. Use of such fermented foods, apart from using locally available materials will provide food acceptable to the local palate. Earlier studies have shown that such fermented foods are more acceptable than high protein biscuits²⁰. Preliminary feeding trials have shown that fermented soy idli incorporated with lime and leafy vegetables is highly acceptable to pre-school children and is well tolerated by children suffering from protein-calorie malnutrition.

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Comparative Studies on Atta (Whole Wheat Flour) and Resultant Atta A By-Product of Roller Flour Milling Industry*

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The resultant atta from roller flour mills contained lesser protein, ether extractives and ash than the whole wheat flour (atta) processed in $chakk^{\dagger}$ (disc mill). The starch damage in whole wheat atta (13.3 to 19.1%) was nearly double than that in resultant atta and indicated greater severity of grindling in chakki. The chapati water absorption as well as Farinograph water absorption and the dough development time were significantly higher in atta. The dough based on resultant atta had more resistance to extension and extensibility than the dough based on atta. The chapatis made from atta were better than those from resultant atta with respect to eating quality as well as flavour.

More than 10 per cent of the 36 million tonnes of wheat produced in India is processed in roller flour mills to obtain refined wheat flour (maida) and semolina (soji) as main products and resultant atta and bran are obtained as by products. The resultant atta, which is also known as 'mill atta' in the trade, accounts for 5-15 per cent of the total milled products and contains different streams of fine bran, shorts, clears and tail fines in various proportions. This by product is cheaper than atta. Resultant atta is generally used for the preparation of chapati in restaurants and industrial canteens, but rarely used in households. Resultant atta was found to be superior to atta nutritionally and it contains about twice as much thiamin and one and half times as much riboflavin and lysine as compared to whole wheat^{1,2}. However, no information is available on the chemical and dough characteristics as well as chapati making quality of resultant atta. Results obtained on these aspects are presented in this paper.

Materials and Methods

Samples (J,M,K,C,W, and F) of resultant *atta* and the wheat grist used to obtain the same, were procured from six different roller flour mills situated in different parts of the country having capacities ranging from 30 to 150 metric tons per day. The wheat grist was milled in a *chakki* (disc mill) to obtain *atta*. The samples of both resultant *atta* and *atta* packed in air tight containers were stored in a cold room (4°C) till used for analysis. *Physical characteristics:* Bulk density of *atta* and resultant *atta* was determined by measuring the volume of 500 g of *atta* in a measuring cylinder. Sieve analysis was carried out using 200 g *atta* samples in a Buhler plane sifter. The overtailing on different sieves after runnnig the sifter for 5 min. were weighed. The colour of the *atta* samples was measured in a Photovolt Reflectance Meter using tristimulus green filter.

Chemical characteristics: Moisture, ash and ether extractives were determined according to standard AACC methods³. Nitrogen content (N) was determined by microkjeldahl method and protein was calculated as $N \times 5.7$. AOAC method⁴ was used to estimate the free fatty acid (FFA).

Rheological characteristics: Water absorption of atta and resultant atta for chapati making was measured using Research Water Absorption Meter as per the method standardised by Haridas Rao et al5. Farinograph characteristics were determined by using Brabender Farinograph⁶ at a chapathi dough consistency of 600 BU and at lever position of 1:3. Extensograph characteistics of chapati dough made in Hobart mixer (model N-50) using water equivalent to chapati water absorption, was determined using Brabender extensograph. Only 75 g. of this chapati dough was used for stretching instead of the usual 150 g used for bread dough, since the chapati dough was stiffer. However, the remaining weight was compensated by keeping 75 g weight along with the dough while stretching. The extensograms were drawn at 0 and 1 hr resting periods, and were evaluated as per the standard AACC method³.

^{*} Presented at the Symposium on 'By Products from Food Industries; Utilization and Disposal' held at Central Food Technological Research Institute, Mysore, on May 29-30, 1980.

Preparation of chapati: Chapati was made from a dough obtained by mixing 100 g atta and resultant atta with pre-determined quantity of water for 3 min. The dough was rolled into a sheet of 2.0 mm thickness with a wooden roller pin on a specially designed aluminium platform and cut into circular shape of 15 cm diameter. The chapati was then baked (each side) on a hot plate maintained at 400°F for 45 sec. followed by puffing in a gas *tandoor* for 10 sec each side. After cooling, the chapati was measured by a panel of 6 judges for its aroma, taste and eating quality. The colour of chapati was measured in a Photovolt reflectance meter using tristimulus green filter.

Results and Discussion

Physical and chemical characteristics of *atta* and resultant *atta* are given in Table 1.

Physical characteristics: Resultant *atta* was more whitish as shown by the higher reflectance value, than the corresponding *atta* which indicates lower bran or higher endosperm content in the resultant *atta*. They were much coarser than *atta* as indicated by the percentage overtailings on a $10 \times \times (130\mu)$ sieve. The higher bulk density of the resultant *atta* may be due to its somewhat uniform particle size higher amounts of denser endosperm and lesser bran fraction.

Chemical characteristics: The moisture content in the resultant atta was 4-5 per cent higher than the atta due to the conditioning of wheat wherein the moisture content of the grain is raised to about 15 per cent. On the contrary, there was a loss of moisture in atta during chakki milling due to frictional heat development. Higher moisture content in the resultant atta may be one of the reasons for its poor keeping quality. Although there was considerable variation in the ash content of different resultant *atta* samples (0.58-1.39 per cent) the values were still consistently lower than the corresponding *atta* samples. This indicated a lower content of bran in the resultant *atta* samples.

Lower fat content in resultant *atta* indicated the presence of lesser amounts of germ, as most of the fat in wheat kernel is concentrated in germ. The presence of low amount of germ was also substantiated by lower protein content, as germ contained about 3 times the protein as compared to endosperm. Higher FFA in resultant *atta* may be due to higher moisture content, which accelerates the hydrolysis of fat.

Physico-chemical characteristics: The damaged starch content in *atta* was more than double as compared to resultant *atta* (Table 1). This indicated higher severity of grinding in disc type grinder (*chakki*) as compared to roller type mill. As damaged starch has higher water absorption capacity than the natural starch, it is logical to expect higher chapati water absorption for *atta*. It was observed that the chapati water absorption increased with the increase in the damaged starch content.

Farinograph characteristics: The farinograms of different resultant atta and the corresponding atta (Fig. 1) showed higher dough development time and dough stability for atta. The excessively higher dough development time may be due to the presence of higher amount of bran particles in atta which may interfere in the quicker development of gluten. Also, preferenctial absorption of water by some constituents other than gluten might also contribute to the longer dough development time.

Extensograph characteristics: The extensibility as

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Characteristics*	-	J	Ν	м	К		С		Ŵ	1	F	
Characteristics	R. atta	Atta	R. atta	Atta	R. atta	Atta	R. atja	Atta	R. atta	Atta	R. atta	Atta
Overtailings of 10X × sieve (%)	71.1	31.3	74.5	23.9	14.7	26.7	78.2	27.7	58.1	27.7	53.4	21.8
Moisture (%)	10.2	8.2	11.1	8.1	9.0	6.0	10.4	7.6	12.4	8.4	13.4	6.6
Ash (%)	0.84	1.55	0.58	1.58	1.24	1.70	1.10	1.50	0.96	1.59	1.39	1.73
Ether extract (%)	1.47	2.10	1.24	1.90	2.12	2.24	1.72	1.96	1.85	3.30	2.46	2.60
Protein (N×5.7) (%)	9.05	9.66	9.05	9.66	9.46	9.92	8.80	9.46	9.27	10.40	10.40	10.65
Free fatty acids (%)	173.6	123.2	210.0	126.0	86.8	81.2	93.2	92.4	78.4	58.8	100.8	67.2
Damaged starch (%)	6.6	14.1	5.9	13.3	10.2	13.9	6.4	19.1	7.4	15.0	6.3	13.7
Chapati water absorption (%)	60.0	69.5	56.5	68.0	61.0	71.0	61.0	72.0	65.0	85.0	60.0	80.0

TABLE 1. PHYSICOCHEMICAL CHARACTERISTICS OF RESULTANT ATTA (R. ATTA) AND ATTA

*Colour values (%) measured in photovolt reflectance meter using tristimulus green filter ranged between 62 and 70 for R. atta and 60 and 64 for Atta. Similarly, density (g/cc) ranged between 0.49 and 0.58 for R. atta and 0.46 and 0.51 for Atta.



FIG. 1. Farinograms of resultant atta (RA) and atta (WMA)

well as the resistance to extension were higher for resultant *atta* as compared to *atta* (Fig. 2). This is possibly due to the presence of higher amount of endosperm fraction, wherein all the gluten forming proteins are concentrated. This is also reflected by strength as indicated by the area of the extensograms. *Atta* forms very stiff dough as shown by higher ratio figure. Though the resistance to extension was higher for dough based on resultant *atta*, at 0 hr, rest period, it was of



FIG. 2. Extensograms of resultant atta (RA) and atta (WMA)

interest to note that on resting for 1 hr, it drastically decreased and was on par with that of *atta* dough, rested for the same period. The higher rate of softening of resultant *atta* during resting was possible due to the presence of high amount of proteolytic enzymes. This indicated that resting the dough made from resultant *atta* improved its rolling characteristics

Chapati making quality: The quality of chapati made from resultant atta and the corresponding atta are given in Table 2. The chapati made from resultant atta was more whitish. The chapati made from resultant atta was more leathery and chewy than that made from atta. This can be attributed to the presence of higher amounts of gluten as the resultant atta contained more of gluten containing endosperm fractions. The extent of leatheriness in chapati made from different resultant atta samples depended generally on their ash content. Chapati made from J. M. and W samples of resultant atta were more leathery as they had very low bran content which is indicated by their low ash contents.

	J	Г	N	М	ŀ	κ	C	;	W		F	
Characteristics	R. atta	Atta	R. atta	Atta	R. atta	Atta	R. atta	Atta	R. atta	Atta	R. atta	Atta
Colour value* (%)	50.0	40.0	57.5	40.5	49.5	41.0	42.0	38.0	45.0	41.5	48.0	44.5
Wheaty aroma	Mild	Good	Mild	Good	Fairly good	Good	Fairly good	Good	Mild	Fairly good	Mild	Fairly good
Chewing Quality	Leathery	Normal	Leathery	Normal	Slightly leathery	Normal	Slightly leathery	Normal	Leathery	Normal	Leathery	Normal
Tase	Bland	Sweet	Bland	Sweet	Slightly sweet	Sweet	Bland	Sweet	Bland	Sweet	Bland	Slightly sweet

*Measured in photovolt reflectance meter using tristimulus green filter.

In contrast chapatis made from K, C and F samples of resultant *atta* were less leathery, as they had comparatively higher bran content. However, very little difference was observed in the eating quality of chapatis made from different *atta* samples showing thereby the similarity in the quality of wheat used in different mills.

The bland taste of chapati made from resultant atta as compared to sweet taste of chapati made from atta was possibly due to the formation of less sugar during resting of the dough, as it contained lesser amount of damaged starch. Austin and Ram⁷ also observed that flour with higher diastatic activity produced sweet chapatis. It is also of interest to note that the typical wheaty aroma observed in chapati made from atta, was not found in chapati made from resultant atta. The better flavour of chapati made from atta is probably due to the development of flavour due to frictional heat produced during grinding in chakki.

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Protein Quality of Chaisathi (Tea Whitener)

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The protein quality of *chaisathi*, a vegetable-protein based milk, was evaluated against that of low fat milk, skim milk and casein. The protein and fat content of *chaisathi* powder were 35 and 18 per cent respectively. Using growth rate, Protein Efficiency Ratio (PER), Biological Value (BV) and Net Protein Utilization (NPU) as the indicators, the *chaisathi* protein was found comporable with that of low fat milk, skim milk and casein. *Chaisathi* seemed to be a good substitute for dairy milk.

Chaisathi is being produced by the National Dairy Development Board (NDDB) of India to provide less expensive milk to socio economicially poor population. It's current production is 3000L/day and is being sold in low income areas at Baroda. Chaisathi contains 10.5 per cent total solids, of which 2.0 per cent is fat. It provides 3.8 per cent protein.

The objective of the present study was to evaluate protein quality of *chaisathi* against protein of low fat milk, skim milk and casein.

Materials and Methods

The samples of *chaisathi*, low fat milk and skim milk powder were obtained from NDDB. Casein was purchased from Amul Dairy, Anand. The protein and fat content of the powders are given in Table 1. The protein evaluation was based on growth rate, protein efficiency ratio, net protein utilization and biological value. The diets based on different sources of materials contained 10 per cent protein (Table 2). For growth rate and PER³ determinations, the male weaning rats

Table 1. protein a and milk	ND FAT CONT BASED SAMP	TENT OF <i>CHAIS.</i> Les	ATHI
Samples		Protein (%)	Fat (%)
Chaisathi		35	18
Low fat milk		35	18
Skim milk powder		40	<1
Casein		92	<1

weighing between 35 and 40g were divided into 4 groups of 10 each and fed different diets for 28 days.

For BV3 and NPU4 determinations, rats weighing between 40 and 50g were divided into 5 groups of 6 each. One group was placed on protein free diet and the remaining groups on experimental diets for a period of 10 days. During the last three days of the experimental period, the rats were placed in metabolic cages to collect urine and fecal material. Urine was collected under toluene. The fecal pellets were collected, air and oven dried and stored until analysis. The rats were ether anaesthetized, the gastrointestinal tract from cardiac to anal end was removed and discarded. The carcasses were weighed, and frozen stored. The urine, fecal and carcass were analysed for nitrogen by microkjeldhal² procedure and protein is calculated by $N \times 6.25$. The carcasses were pressure cooked and the homogenate was prepared according to the method of Mickelson et al.⁵ for nitrogen determination. NPU and BV of the protein were calculated.

Student's 't' test was used to find out significant differences between the two means. All tests were considered at 95 per cent of significance⁶.

Results and Discussion

The daily food intake and consequently weight gain did not appreciably vary among the groups although

TABLE 2.	COMPOSITI	ON OF TH	E EXPER	IMENTAL	DIETS
Ingredients	Casein	Chaisathi	Low fat milk	Skim milk	Protein free
	(g)	(g)	(g)	(g)	(g)
Protein source	11	29.0	29.0	25	0
Vitamin mix ¹	2	2.0	2.0	2	2
Salt mix ²	4	4.0	4.0	4	4
Groundnut oil	10	9.5	9.5	10	10
Sago	73	55.5	55.5	59	84

 TABLE 3.
 MEANS AND STANDARD ERROR: FOR FOOD INTAKE, GROWTH

 RATE AND FOOD UTILIZATION

Groups	Food intake (g/day) Mean±S.E.	Wt. gain (3/day) Mean±S.E.	Food intake/wt (g) gain Mean±S.E.
Chaisathi fed	12.05±0.220	3.11±0.187	3.98±0.217
Low fat milk fed	12.14±0.325	3.12±0.146	3.95±0.179
Skim milk fed	13.09±0.242	3.21±0.116	4.50±0.399
Casein fed	12.43 ± 0.347	3. 7 0±0.247	3.44±0.159

TABLE 4. PER, NPU AND BV OF CHAISATHI AND OTHER MILK BASED SAMPLES

	0.000		
Source	PER Mean \pm S.E.	NPU Mean±S.E.	BV Mean±S.E.
Chaisathi	2.58 <u>+</u> 0.146	$81\pm$ 4.80	73±2.081
Low fat milk	2.58±0.126	$76\pm$ 5.406	69±1.995
Skim milk	2.45±0.092	86±11.108	84±3.602
Casein	2.96±0.147	80± 6.929	81±3.921

casein fed group tended to gain relatively more weight (Table 3). The *chaisathi* fed group was as efficient in food utilization as the low fat milk and casein fed groups. The former tended to be more efficient in food utilization than the skim milk protein fed group(Table 3).

The mean value for PER of *chaisathi* protein was comparable to those of low fat milk and skim milk proteins (Table 4). Likewise, the mean values for BV and NPU of different proteins did not vary appreciably. The value for NPU of protein of *chaisathi* was quite comparable with that of casein protein, while the BV of *chaisathi* protein was about 10 per cent lower than that of casein. Based on its PER, NPU and BV, it appears that *chaisathi* could be a good substitute for dairy milk.

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Studies on Pectin Yield and Quality of Some Guava Cultivars in Relation to Cropping Season and Fruit Maturity

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Five guava cultivars ('Banarasi Surkha', 'Allahabad Safeda', 'Tehsildar', Apple Colour' and 'Sardar') were evaluated for pectin contents as affected by the cropping season (rainy and winter) and stage of fruit maturity, (green mature, unripe and half ripe). Significantly higher amounts of crude pectin and jelly units were observed in winter season fruits but the pectin quality in terms of methoxyl content, anhydrogalacturonic acid content, degree of esterification and jelly grade was better in rainy season fruits. Although stage of fruit maturity had no marked effect, yet half ripe fruits yielded more jelly units than unripe ones. "Banarasi Surkha' and 'Sardar' yielded high grade pectin. but higher number of jelly units were obtained from 'Sardar' and Apple Colour'.

Guava fruit has been found to be a rich source of pectin¹. Various factors like stage of fruit maturity¹, cropping season² and type of cultivar¹⁻⁴ affected the quantity and quality of pectin extracted from the tissue. At present, production of pectin in India is quite short of the requirements. Guava is fairly rich in pectic substances and its cultivation has bright scope in the arid zones of North India. Present study was therefore, undertaken to find out the effect of cropping season and appropriate stage of fruit maturity for maximum pectin extraction from important guava cultivars.

Materials and Methods

Five guava cultivars 'Banarasi Surkha', 'Allahabad Safeda,' 'Tehsildar', 'Apple Colour' and 'Sardar' (Syn. 'Lucknow-49') were evaluated for their pectin content. Both rainy and winter season crops during the year 1978-79 were included in the study. Fruits were harvested at two stages: (i) Green mature unripe, and (ii) half ripe. To ascertain the proper stage of harvest, the firmness of guava fruits was determined with the Magness Taylor pressure type tester, fabricated at the Department of Agricultural Engineering, HAU, Hissar. The fruits with more than 8.5 kg/cm² pressure were considered as green mature unripe during both the cropping seasons. Depending on the cultivar, fruits with pressure range of 5.108 to 6.479 kg/cm² during rainy season and of 5.460 to 6.628 kg/cm² during winter season were considered as half ripe.

Preparation of Sample for Pectin Extraction: Atter thorough washing and surface drying, the fruits were

passed through an electric grater and whole of the crushed material was homogenised. For extraction and isolation, the procedure described by Mc-Cready⁵ was adopted with the following details. Freshly grated guava pulp (500 g) was boiled in 1.0 l. water. The pH of the slurry was maintained at 4.0. It was heated for 35 min. at 95-100°C while being stirred continuously. It was then filtered through muslin cloth and immediately cooled below 25°C. The isolation of pectin was done with 70 per cent iso-propanol containing 0.01 N HCl, followed by four washings with 95 per cent isopropanol. The precipitates of pectin, thus obtained were dried at 35°C in forced draft oven for about 24 hr, till constant weight was obtained. The dried precipitates were weighed and per cent crude pectin yield was calculated. The dried light brown pectin was ground to pass a 60 mesh screen for further analysis.

Characterisation of pectin. The moisture content was determined by drying 1g of pectin at 90°C to a constant weight and the ash content was estimated by ashing in a muffle furnance at 550-600°C for 5 hr.⁶ The equivalent weight, methoxyl content and anhydrogalacturonic acid content were determined by the methods of Owens *et al.*⁷ Degree of esterification was calculated on the basis of methoxyl and anhydrogalacturonic acid contents⁸. The jelly grade of the pectin was determined by relative viscosity method⁹ and the jelly units were calculated by multiplying the jelly grade, with the crude pectin yield obtained from 500 g of fresh guava fruits.

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Results and Discussion

The winter season guavas yielded significantly higher pectin than the rainy season fruits (Table 1), which may be attributed to low moisture content and more compact cells in winter fruits². Presence of higher pectin in winter fruits has also been reported by Sachan *et al*¹⁰ and Gangwar¹¹. Although half ripe fruits gave better pectin yields than the unripe ones, the differences were not significant. Different cultivars did not vary significantly in their pectin yields. However, 'Apple Colour' during winter season and 'Tehsildar' during rainy season gave relatively higher pectin yield.

No particular trend was noted regarding moisture and ash contents of the extracted pectin. In general, pectin from winter guavas had higher moisture content than from rainy season (Table 1). The ash content showed no impact of either cropping season or stage of maturity.

Data given in Table 2 reveal that cropping season had no significant effect on the equivalent weight of the pectin. Further, the half ripe fruit yielded pectin with higher equivalent weight, but the differences were not significant. Among the cultivars, 'Banarasi Surkha' and 'Tehsildar' during rainy season and 'Apple Colour' and 'Tehsildar' during winter season gave pectins with the highest equivalent weights. Pectins from rainy season fruits had significantly higher methoxyl content than those from winter crop (Table 2), but the stage of fruit maturity showed no significant effect. Among the cultivars, 'Banarasi Surkha' during rainy season and 'Apple Colour' during winter season yielded pectins with comparatively higher degree of methoxylation.

The anhydrogalacturonic acid (AUA) content and degree of esterification (DE) which indicate the purity and percentage of total uronide carboxyl groups that are esterified with methanol, respectively, were found to be higher in pectin from rainy season guavas (Table 2). The advancement of maturity had no significant effect on AUA or DE, although the former, showed a decreasing trend. The pectin from 'Banarasi Surkha' contained the highest AUA content and also the DE.

Irrespective of all the above factors, rainy season fruits yielded pectins with significantly higher jelly grade over winter season fruits (Table 3). The half ripe fruits yielded slightly higher grade pectin than the unripe ones. Among cultivars, 'Banarasi Surkha' and 'Sardar' yielded high grade pectins, while 'Tehsildar' had the lowest grade pectin. The jelly grade was found to be inversely proportional to the crude pectin yield(Table 1), but it was directly proportional to the methoxyl content (Table 2) which is an index of esterification and an

		Crude	pectin yiel	d (%)	Mo	oisture (%)		Ash (%)	
Cultivar	Crop	Green mature ur.ripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean
Banarasi	Rainy	0.71	0.77	0.74	10.7	9.4	10.1	1.2	1.1	1.2
Surkha	Winter	0.79	1.03	0.91	12.7	12.4	12.6	1.3	1.6	1.5
Allahabad	Rainy	0.74	0.77	0.76	13.4	10.5	12.0	1.2	1.1	1.2
Safeda	Winter	0.93	0.96	0.95	14.9	12.2	13.6	1.7	1.9	1.8
Tehsildar	Rainy	0.66	0.91	0.79	9.8	10.1	10.0	1.2	1.4	1.3
	Winter	0.97	1.00	0.99	11.8	10.4	11.1	1.3	1.6	1.5
Apple	Rainy	0.69	0.87	0.78	14.4	12.4	13.4	1.7	1.9	1.8
Colour	Winter	1.08	1.34	1.21	12.3	14.7	13.5	1.6	1.8	1.7
Sardar	Rainy	0.74	0.79	0.77	12.1	11.7	11.9	1.4	1.7	1.6
	Winter	0.90	1.05	0.98	10.2	11.0	10.6	1.5	1.8	1.7
Mean	Rainy	0.71	0.82	0.77	12.1	10.8	11.5	1.3	1.4	1.4
	Winter	0.93	1.08	1.01	12.4	12.1	12.3	1.5	1.7	1.6

 TABLE 1.
 EFFECT OF CROPPING SEASON AND STAGE OF FRUIT MATURITY ON CRUDE PECTIN YIELD, MOISTURE AND ASH CONTENTS OF PECTIN

 IN VARIOUS GUAVA CULTIVARS

't' at 5%: Difference between stages of maturity is not significant for crude pectin yield, moisture and ash. Difference between seasons is significant for crude pectin yield only.

Cultiver	Gran	Equiv	alent w	eight	Meth (Ash-H	oxyl (M 2 O –free	e0%) basis)	Anhyđ	rogalact acid (%)	uronic)	Degree	of ester (%)	ification
Cuntvar	Стор	Green mature unripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean
Banarasi	Rainy	834.5	830.1	832.3	7.69	7.52	7.61	69.24	70.40	69.82	63.11	60.95	62.03
Surkha	Winter	674.3	740.0	707.2	5.83	6.08	5.96	59.02	61.43	60.23	56.10	56.13	56.12
Allahabad	Rainy	763.3	715.1	739.2	6.51	6.61	6.56	62.18	66.21	64.20	59.32	56.67	58.00
Safeda	Winter	700.5	720.3	710.4	6.46	5.06	5.76	64.07	56.25	60.16	57.22	51.09	54.16
Tehsildar	Rainy	747.8	837.5	792.7	5.76	6.15	5.95	59.34	59.01	5 9 .18	55.08	59.17	57.13
	Winter	829.8	810.7	820.3	5.63	6.14	5.8 9	56.31	59.70	58.01	56.76	58.35	57.56
Apple	Rainy	639.8	785.0	712.4	7.50	6.20	6.85	72.26	59.70	65. 9 8	58.90	58.89	58.90
Colour	Winter	740.0	851.0	795.5	6.14	6.19	6.17	61.77	59.02	60.40	56.39	59.54	57.97
Sardar	Rainy	646.8	713.3	680.1	7.62	6.73	7.18	72.64	65.87	69.26	59.53	57.96	58.75
	Winter	756.4	767.4	761.9	5.70	5.43	5.57	58.67	56.97	57.82	55.10	54.17	54.64
Mean	Rainy	726.4	776.2	751.3	7.02	6.64	6.83	67.13	64.24	65.69	59.19	58.73	58.96
	Winter	740.2	777. 9	759.1	5.95	5.78	5.87	59.97	58.67	59.32	56.31	55.86	56. 09
't'at 5%	Stages	No	ot signifi	cant	No	ot signifi	cant	N	ot signif	icant	No	t signific	ant
	Seasons	No	t signifi	cant	:	Significa	nt		Significa	ant	S	Significan	t

TABLE 2. EFFECT OF CROPPING SEASON AND STAGE OF FRUIT MATURITY ON BIOCHEMICAL CHARACTERS OF GUAVA PECTIN

TABLE 3. EFFECT OF CROPPING SEASON AND STAGE OF FRUIT MATURITY ON JELLY GRADE AND JELLY UNITS OF GUAVA PECTIN

			Jelly grade		Jelly units/500g fruit)			
Cultivar	Сгор	Green mature unripe	Half ripe	Mean	Green mature unripe	Half ripe	Mean	
Banarasi Surkha	Rainy	184	206	195	650	778	714	
	Winter	157	159	158	622	817	720	
Allahabad Safeda	Rainy	176	162	169	655	623	639	
	Winter	146	151	149	666	723	695	
Tehsildar	Rainy	152	158	155	501	721	611	
	Winter	144	147	146	7(4	734	719	
Apple Colour	Rainy	18 9	172	181	651	744	698	
	Winter	129	135	132	697	906	802	
Sardar	Rainy	190	200	195	697	795	746	
	Winter	147	149	148	662	780	721	
Mean	Rainy	178	180	179	631	732	682	
	Winter	145	148	147	6 7 0	792	731	
't' at 5%	Stages	Ν	lot significant		Rainy sea	son : Not si	ngificant	
	Seasons		Significant		Winter se No	ason : Signific t significant	cant	

important factor in controlling the setting time of jellies.

Winter season fruits yielded more jelly units than the rainy season ones. In rainy season, the stage of fruit maturity did not significantly affect the jelly units, whereas in winter crop, the half ripe fruits yielded significantly higher jelly units than the unripe ones. Among the cultivars, 'Sardar' yielded pectins with the highest jelly units in rainy season, whereas, in winter season, the highest number of jelly units were recorded in 'Apple Colour', followed by 'Sardar'. The lowest jelly units in both the cropping seasons were obtained in pectin from. 'Tehsildar', Although the pectin from 'Banarasi Surkha' had the highest jelly grade, yet due to lesser pectin yields, the jelly units were comparatively lower than those from 'Apple Colour' guavas.

From these results, it is clear that the rainy season crop y elded pectin with high jelly grade and higher degree of methoxylation, but due to low yields, the jelly units were lower while winter crop pectin had low jelly grade and lower methoxyl content. Due to higher pectin yields, the number of jelly units was found higher. Although the stage of fruit maturity had no significant effect, yet half ripe fruits gave more pectin yield. Among the cultivars, 'Apple Colour', 'Sardar' and 'Tehsildar' were found promising with respect to pectin yield and jelly units.

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Penetration of Bacteria into Muscles of Goat, Pork and Poultry Meat

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Penetration of four pathogenic bacteria into goat, pork and poultry meat was studied. Bacillus cereus and Staphylococcus aureus caused linear proteolysis, while Salmonella typhimurium penetrated along the perimyseal septa in goat muscle. In pork muscle, B. cereus and Escherichia coli showed linear proteolysis, while S. aureus and S. typhimurium produced extensive liquifaction along the peri-and endomyseal septa. In poultry muscle, proteolysis occurred along peri-and endomyseal septa by B. cereus, S. aureus and E. coli.

The spoilage of meat by proteolytic bacteria is important not only from meat hygiene, but also from public health viewpoint as these bacteria involved in the process are usually pathogenic or toxigenic in nature. Bottom *et al.*¹ and Hesengawa *et al.*² observed that muscle protein breakdown occurred during microbial

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spoilage of meat. Tarrant *et al.*³ had shown that the onset of spoilage in pork by *P. fragi* coincided with extracellular proteolytic activity and an increased amount of ultrastructure damage to muscle fragments. In pure culture study in beef meat, Dainty *et al.*⁴ reported that proteolytic spoilage was mainly due to gram negative bacteria. Penetration of bacteria into meat is caused by proteolytic species only. Thus non-proteolytic bacteria did not invade even when associated with proteolytic organisms.⁵

The present investigation has been aimed to determine the extent to which food poisoning bacteria invade the deeper layers of different types of meat. This study is expected to strengthen the understanding of the spoilage process of meat.

Materials and Methods

Cultures: The bacterial cultures used were Bacillus cereus, Staphylococcus aureus, Escherichia coli and Salmonella typhimurium. These cultures were procured from Central Research Institute, Kasauli (India). All cultures were maintained on nutrient agar slants and were grown on nutrient agar plates.

Collection of samples: The meat samples were collected aseptically from the slaughter house between 6 and 7 a.m. and analysed within 2 hr. The meat samples were normal and pink in colour.

Penetration of bacteria: The muscles were cut into strips of meat $2 \times 2 \times 5$ cm in size in case of goat and pork and $2 \times 2 \times 4$ cm in case of poultry. Meat samples were obtained from thigh portions of the goat and pork and sternal muscles of the poultry. They were dipped in alcohol to remove the surface microflora⁶. The meat strips were immersed in 3 per cent molten agar containing 0.1 per cent mercuric chloride to inhibit the growth of surface contaminants. The solid agar was removed from one end of the meat strip and inoculated with the bacteria. Two strips of meat were inoculated at the same time for each experiment. These inoculated strips of meat were hooked to the pins hanging in a stand placed in the jar. This ensured the incubation of the meat strips vertically. The jar containing the strips was placed at 37°C until the organisms had shown sufficient growth, i.e. for 20 hr in case of goat and poultry meat for all the types of bacteria studied; 44 hr in case of pork for B. cereus and S. aureus and 20 hr for E. coli and S. typhimurium. The trials were conducted in duplicate.

After the incubation, one strip each of the meat inoculated with different bacteria was placed in a tube containing sufficient amount of 10 per cent formalinsaline. The tube was placed in the water bath at 56°C for 30 min. to fix the tissue. Pieces of 0.3 cm were removed from the inoculated end from a distance of 2, 3 and 4 cm of the strips in case of goat and pork and 2 and 3 cm in case of poultry meat. The sections were cut by embedding in paraffin wax, stained with hemotoxylin eosin stain.

Another strip of meat was cut at 2, 3 and 4 cm from the inoculated end and streaked on the selective media and incubated at 37°C for 24-28 hr. Colonies from the plates showing growth were further examined to confirm the identity of the organism.

Selective media: Mannitol egg-yolk phenol polymyxin agar for *B. cereus, Staphylococcus* medium 110 for *S. aureus*, brilliant green agar for *E. coli* and Salmonetla-Shigella agar and desoxycholate citrate agar for *S. typhmiurium* were used.

Results

Goat muscle strips: B. cereus and S. aureus penetrated each to a depth of 3 cm producing linear proteolys's by liquefying the muscle fibres and their surrounding septa, whereas S. typhimurium penetrated along the perimyseal septa to a depth of 4 cm. E coli did not show any indication of penetration (Table 1, Fig. la and lb). The muscle strips were incubated for 20 hr when the organism had attained maximum growth.

TABLE 1.	PENETRATION C	OF VARIOUS	BACTFRIA	IN	MUSCLES	OF
	DIFFEREN	T SPECIES OF	MEAT ANIM	ALS	5	

Source of meat	of Bacteria	Incubation period (hr)	Depth of penetration* (cm)	Type of proteolysis
Goat	B. cereus	20	3	Linear
	S. aureus	20	3	Linear
	E. coli	20		No proteolysis
	S. typhimurium	20	4	Along muscular septa
Poultry	B. cereus	24	3	Along muscular septa
	S. aureus	24	3	Around the muscle fibres
	E. coli	24	3	Along muscular septa in bran- ching manner & in muscle bundles
	S. typhimurium	24	3	Along muscular septa and linear
Pork	B. cereus	44	4	Linear
	S. aureus	44	4	Around muscle fibres
	E. coli	20	4	Linear
	S. typhimurium	20	4	Linear

*As evidenced by proteolysis and culturing.



FIG. 1a. Goat Muscle strip infected with B. cereus: linear proteolysis H. E. X 100.



FIG. 1b. Goat muscle strip infected with S. typhimurium: proteolysis along the muscular septa H. E. X 100.

Pork muscle strips: All the organisms showed a penetration upto 4 cm (Table 1, Fig. 2). Maximum growth was shown by B. cereus and S. aureus in 44 hr and E. coli and S. typhimurium in 20 hr. B. cereus and E. coli produced linear proteolysis dissolving the muscle fibres as well as the encircling endomyseal septa, whereas S. aureus and S. typhimurium produced extensive liquifaction along the peri-and endomyseal septa usually sparing the muscle fibres.



FIG. 2. Pork muscle strip infected with S. aureus: proteolysis around the muscle fibres H, E. X 100.



FIG. 3. Poultry muscle strip infected with *B. cereus*: proteolysis along muscular septa and extending around the individual muscle fibres H. E. X 100.

Poultry muscle strips: Each of these organisms penetrated to a distance of 3 cm along the muscular septa within 24 hr (Table 1 Fig. 3). The proteolysis in *B. cereus*, *S. aureus* and *E. coli* extended along the periand endomyseal septa, whereas in *S. typhimurium* the proteolysis was noticed only along the perimyseal septa.

Discussion

The penetration of meat by bacteria results from the breakdown of the muscle fibres as well as the associated connective tissue in the interstitial septae. The pattern of proteolysis varied depending upon the type as well as the animal species from which the muscle strips were derived.

In goat muscle strip, the linear proteolysis caused by *B. cereus* and *S. aureus* as compared to the proteolysis by *S. typhimurium* along the perimyseal septa indicate

that the former two organisms had comparatively more proteolytic activity as the linear penetration occurs due to the dissolution of muscle fibres as well as the peri-and endomysium. In pork muscle, the linear liquefaction by B. cereus and E. coli as compared to the perimyseal dissolution in case of other two organisms likewise indicate high degree of lytic activity in case of the former two organisms. In poultry strips, the proteolysis by B. cereus, S. aureus and E. coli extended along the periand endomyseal, whereas it was only along the perimyseal septa in S. typhimurium suggesting the higher proteolytic activity in case of former three organisms. The deeper penetration in case of pork as compared to goat muscle indicated, that the pork muscle fibres were easily penetrable which might be due to relatively soft nature of the muscle fibres. The strip of the poultry muscle was only 3 cm in length and the penetration of the organism was observed throughout the strip, the chances of deeper penetration in the event of the use of the longer strip could not be ruled out. The variation in the penetration of bacteria in muscles of goat, poultry and pork reflected variation in the nature and extent of production of proteases. This could possibly be due to the variable nutritional conditions provided by muscle strips of different species. The penetration had been reported to occur only due to proteases and the non-proteolytic species did not invade, even when they were present along with proteolytic species. This was probably because the penetration originated in the area of growth of a microcolony of the proteolytic species and thus nonproteolytic bacter'a were excluded⁵.

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Fractionation of Wood Phenolics and Their Use in Brandy

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Phenolics were extracted from Indian oak (Quercus sp.), teak (Tectona grandis), red cedar (Cedrella toona) and sandal (Santanun album) by ethanol. Parameters like ratio of wood to ethanol, concentration of solvent, and time of contact for extraction of phenolics were studied. Extracted phenolics were segregated into, flavanoids and non-flavanoids as well as soluble and insoluble fractions in solvents like chloroform, ethylacetate and amyl alcohol. The fractionated phenolics were estimated after removal of the solvents. Studies on organoleptic evaluation of young brandy on addition of these fractions of phenolics (free from solvents) have indicated that some combinations of fractions reduce the harshness of the product.

It is an age old practice to store wine and distilled liquor such as brandy, whisky and rum in wooden casks. The type of woods used for casks are specific and they impart desirable characteristics to the product. The changes that take place during storage or "ageing" or "maturation" as it is called, is not very well understood. Apart from the interaction among the different constituents of the content, the phenolics extracted from wood have an important role in the formation of compounds which are directly responsible for the changes. Singleton¹ used ethanol extract and chips of oak wood to impart oak flavour to wine. Development of aromatic flavour in brandy treated with Indian oak wood chips was reported by Venkataramu *et al*². In the present study, data on fractionation of phenolics from four Indian woods with different solvents in single stage and

also sequentially extracted, along with extraction of flavanoids, and non-flavanoids and phenolics extracted by other solvents are presented. Effect of addition of some of these extracted phenolics on the taste of brandy is also recorded.

Materials and Methods

Teak (Tectona grandis) and red cedar (Cedrella toona) woods were obtained from local timber yard. Deoiled sandal wood (Santanum album) was obtained from Sancal Wood Oil Factory, Mysore. Oak wood (Ouercus species) was obtained from Forest Department of Himachal Pradesh. These woods were powdered and used.

Distilled solvents such as ethanol, ethyl acetate, chloroform, n-butanol and amyl alcohol and AR grade formaldehyde, gallic acid and trichloroacetic acid were used.

Total phenolics were estimated using colorimetric procedure of Singleton and Rossi³ using spectronic 20 (gallic acid was used as standard). Flavanoids and nonflavanoids were determined by the procedure of Kremling and Singleton⁴.

Extraction of phenolics was done using different concentrations of ethanol by stationary and under agitation (Emenvee rotary shaker, 230 r.p.m) and samples were drawn at intervals of 24 hr for analysis. In case of sequential extraction, wood powder was completely submerged in ethanol and every 24 hr, the supernatant was removed and was replaced by fresh ethanol. Addition of formaldehyde and estimation of phenolics in the supernatant gave the value for non-flavanoids. The difference between the total flavanoids and nonflavanoids was computed and the content of flavanoids was calculated. In case of immisible solvent system, the phenolics in different solvents were estimated after separating them.

Results and Discussion

The quantity of phenolics extracted by 45-60 per cent ethanol varied from 4 to 6 per cent by weight of powdered Indian oak, teak, red cedar and sandal woods, at each intervals of 24 hr, under agitation (Table 1).

The quantity of phenolics extracted per gram of wood was highest in red cedar followed by Indian oak, sandal and teak. In red cedar, the ethanol extractable phenolics formed nearly 28 per cent by weight of wood. This appears to be on the high side. It is not known whether this high value is due to any artefact introduced by any other constituent of wood which possess a similar reaction as phenolics with the reagent. The extraction of phenolics appeared to be almost complete even at the end of the first 24 hr. as there was no significant increase in the further extraction at the end of 72 hr.

Singleton et al.⁵ have suggested that flavanoids content in wine matured in wooden casks can be a measure of maturation. Differentiation of flavanoids from nonflavanoids based on their solubility in formaldehyde is one of the recognised methods. This method was used to measure the formaldehyde soluble (non-flavanoids) and formaldehyde insoluble (flavanoids) in these woods.

The results presented in Table 2 indicate that the red cedar contained a high amount of flavanoids followed

Concn.	Concu.]	Indian oak			Teak		Red c	edar*		Sandal	
(%)	alcohol (%)	24 hr	48 hr	72 hr	24 hr	48 hr	72 hr	24 hr	72 hr	24 hr	48 hr	72 hr
4	45	146.5	146.5	155.0	24.9	23.5	23.5	269.2	271.9	56.75	61.25	67.50
,,	50	182.5	182.5	197.5	25.0	26.0	24.5	261.0	271.9	53.75	58.75	66.25
,,	55	210.0	210.0	210.0	30.5	26.5	26.0	269.2	271.9	61.25	73.75	67.50
	60	146.5	146.5	153.7	29.0	27.5	26.0	264.8	271.9	58.75	58.75	67.50
5	45	158.0	158.0	158.0	24.0	24.4	23.6	215.3	282.7	53.00	54.00	56.00
,,	50	182.0	182.0	188.0	25.6	25.2	23.6	215.3	282.7	53.00	54.00	60.00
.,	55	124.0	124.0	136.0	26.0	26.0	24.4	226.2	261.0	56.00	58.00	64.00
	60	152.0	158.0	164.0	26.0	26.0	25.6	212.1	261.0	56.00	58.00	64.00
6	45	128.3			22.0	23.0	22.6	199.0	262.8	48.33	56.66	57.49
,,	50	93.3	100.0	103.3	23.3	23.3	23.3	199.0	253.7	56.66	57.49	59.99
,,	55	121.6	146.6	126.6	25.6	25.5	24.3	189.8	262.8	53.33	56.66	57.49
	60	150.0	155.0	155.0	25.6	24.5	24.6	195.9	271.1	54.99	59.99	59.99

*Not estimated for 48 hr.

TABLE 1

	1s	t extract		2n	d extract		- 3rc	3rd extract			
Type of Wood	Initial phenolics (mg/g of wood)	Flavanoids (mg)	Non Flavanoids (mg)	Initial phenolics (mg/g of wood)	Flavanoids (mg)	Non Flavanoids (mg)	Initial phenolics (mg/g of wood)	Flavanoids (mg)	s Non Flavanoids (mg)		
Indian oak	195.0	189.0	6.0	63.0	58.1	5.0	6.8	5.5	1.3		
Teak	27.0	19.7	7.3	26.0	21.1	4.9	2.8	2.4	0.4		
Red cedar	271.9	263.7	8.1	111.6	107.1	4.4	29.7	28.5	1.2		
Sandal	60.0	57.3	2.7	10.4	8.2	2.2		_			
*50% Ethanol	extraction after	24 hr.									

TABLE 2. FLAVANOIDS AND NON-FLAVANOIDS PRESENT IN INDIAN OAK, TEAK, RED CEDAR AND SANDAL WOODS*

TABLE 3. FRACTIONATION OF ETHANOL SOLUBLE PHENOLICS USING CHLOROFORM, AMYL ALCOHOL AND ETHYL ACETATE

Turne of wood	Ethanol	Chlo	oroform	Amyl	alcohol	Ethyl acetate		
	phenolics (mg/g of wood)	Soluble (mg)	Insoluble (mg)	Soluble (mg)	Inscluble (mg)	Soluble (mg)	Insoluble (mg)	
Indian oak	182.0	23.8	152.6	141.2	37.9	54.9	122.4	
Teak	25.6	5.0	15.4	17.1	6.7	9.8	14.3	
Red cedar	272.7	23.9	246.5	171.6	99.7	139.3	127.2	
Sandal	56.0	22.1	28.2	36.5	17.1	28.2	23.3	

by Indian oak, sandal and teak as observed earlier. These data indicate that flavanoid content in red cedar was significantly high compared to the other woods analysed. Sequential extraction simulates conditions of extraction that would occur in reused barrels. The trend in concentration of flavanoids indicates that repeated use of barrels may extend the period of ageing. From economic considerations it would seem preferable to transfer liquor in a few days from a new barrel to an used one; or the residence time of contents be controlled by frequent analysis to allow dissolution of desirable level of phenolics.

The ethanol extractable phenolics of four woods were fractionated using different solvents such as chloroform, amyl alcohol, and ethyl acetate. The concentration of soluble and insoluble phenolics of these woods in these solvents are presented in Table 3.

The data indicate a general trend that the chloroform insoluble is more than soluble phenolics. Aqueous soluble phenolics is more in sandal followed by teak, Indian oak and lowest being in red cedar. The trend reverses when amyl alcohol is used, where solvent solubles are more compared to insolubles. Aqueous soluble phenolics was highest in Indian oak followed by teak, sandal and lowest in red cedar. In case of ethyl acetate the aqueous layer had high phenolics in Indian oak and teak, while phenolics in solvent layer was high when red cedar and sandal woods were used.

The phenolic fractions of the same wood either singly or in combination were added to fresh/green brandy and subjected to sensory evaluation along with controls. Results indicated that combination of phenolic fractions of the wood when added improved the taste of the brandy.

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The Microbiological Quality of Ice Creams Sold in Hyderabad City

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The bacteriological quality of ice cream and ice fruits sold in Hyderabad were assessed. Three hundred and one samples were collected from hotels, restaurants, ice cream parlours, and local vendors. These were assessed for standard plate count (SPC) coliforms by MPN method and for pathogens. Only 58 (19.4%) were within ISI specifications for SPC and coliform counts. *Staphylococcus aureus* was isolated from 20.3% samples, and *Escherichia colli* from 7% samples. The slab ice cream sold in hotels was superior to other varieties sold by local vendors.

Currently available data suggest that the foods sold in Indian markets are of very poor quality¹⁻³. Ice cream is a popular milk product consumed by people of all categories. The present study was undertaken to assess the microbiological quality of the ice creams sold in Hyderabad city (India).

Materials and Methods

Three hundred and one samples of ice creams consisting of slabs (31), cups (70), bars (25), and lollies (150) were collected in sterile, covered containers from hotels, restaurants, parlours and local vendors. They were brought to the laboratory in ice flasks and immediately analysed.

Analysis: The samples were allowed to thaw and graded dilutions were prepared using Ringer's solution. The total counts were estimated by the standard method⁴. The coliforms were estimated by the most probable number method using MacConkey's broth⁵. Gas positive tubes were streaked on EMB agar and colonies showing greenish metallic sheen were further confirmed by biochemical reactions as *Escherichia coli*.

Staphylococci were isolated by inoculating the sample into mannitol-salt broth for enrichment, followed by streaking on milk-salt agar for isolation. The Staphylococci strains obtained thus were tested for their coagulase activity and confirmed as S. aureus.

Salmonella was tested according to the method of Thatcher and Clark⁶. Modified bile agar medium was used for the isolation of the vibrios, after processing through alkaline peptone water for enrichment⁷. Potassium tellurite medium was used for isolation of diphtheria bacillus.

The enterococci were isolated by inoculating the

sample into SF medium and then streaking on sodium azide medium.

The identification of isolated organisms, were confirmed by biochemical reactions,^{8,9} wherever necessary.

Results and Discussion

The ISI has prescribed SPC of not more than 250000/g and coliform count of not more than 100/g. and the ice cream should be free from pathogenic organisms¹⁰. Out of the 301 samples studied, only 58 (19.4 per cent) were within the standard limits (Table 1). There was a definite correlation between the cost of the ice cream and its sanitary quality. The costlier slab ice creams (77.4 per cent) sold in the hotels were within the standard limits, when standard plate counts and coliform counts were considered together. Among the other varieties, only 12.9 per cent of the cup ice creams, 24 per cent of bar ice creams, 9 per cent of the cones, and 10.7 per cent of the lollies were within the limit specified by ISI.

Of the samples analysed, 88.7 per cent were positive for coliforms. The coliforms isolated included *Klebsiella*, *Enterobacter cloaca*, *E. aerogenes* and *E. hafniae*. However, only 19.3 per cent had counts higher than ISI specifications. *Escherichia coli* was found in 7 per cent of the samples.

Enterococci were isolated from 44.7 per cent of the samples. Since raw milk and dairy products contain *Enterococci* as contaminants, their presence alone does not indicate the sanitary history of the product, unless associated with other indicator bacteria.

Staphylococcus aureus was the only pathogen isolated. This was more frequently isolated from the cheaper varieties of ice creams where 20.3 per cent of the samples were positive for the pathogen (Table 2).

_	St	andard plate cou	nts	Coliforr	n counts	No. within	
Ice cream type	Samples with $<25\times10/4$ g	Samples with $>25\times10^4/g$	Average count (log)	Samples with <100/g	Samples with >100 g	ISI	
Slab	25 (80.6)	6 (19.4)	5.106	29 (93.5)	2 (6.5)	24 (77.4)	
Cups	10 (14.3)	60 (85.7)	5.597	64 (91.4)	6 (8.6)	9 (12.9)	
Chacobars	9 (36.0)	16 (64.0)	5.545	21 (84.0)	4 (16.0)	6 (24.0)	
Cones	5 (20.0)	20 (80.0)	6.572	18 (72.0)	7 (28.0)	3 (9.0)	
Lolly	20 (13.3)	130 (86.7)	5.793	111 (74.0)	39 (26.0)	16 (10.7)	
Total	69 (22.2)	232 (77.8)		243 (80.7)	58 (19.3)	58 (19.3)	

TABLE 1. STANDARD PLATE COUNTS AND COLIFORM COUNTS IN ICE CREAM SAMPLES

Figures in the parenthesis indicate the percentages.

TABLE 2. PATHOGENS ISOLATED FROM THE ICE CREAM

		Staphyl	ococcus				
Ice cream	Samples	aur	eus	Escherichia co l i			
type	collected	No. of samples	% samples	No. of samples	% samples		
Slab	31	5	16.1	1	3.2		
Cups	70	14	20.0	I	1.4		
Cones	25	5	20.0	I	4.0		
Chacobar	25	12	48.0	2	8.0		
Lolly	150	25	16.8	16	10.7		

The poor quality of the cups, cones and wrappers used is probably one of the factors contributing to the heavy load of organisms. Repeated handling and unhygienic sorroundings is another factor for the poor quality of the cheap, locally sold ice creams. Small scale manufacturers produce low grade ice creams of substandard quality. The consumption of these ice creams is an important source of infection.

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Studies on Sunflower Oil with Reference to its Keeping Quality

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Ten sunflower oil samples of five different varieties from irrigated and rainfed trials were studied for their keeping quality. Samples were withdrawn at every thirtieth day and different chemical tests were carried out. Different varieties behaved differently. The keeping quality of the oils obtained from crop grown under rainfed conditions was better than those obtained from irrigated condition.

Sunflower oil is valued highly as a source of polyunsaturated fatty acids. It has a high level (60-72 per cent) of the essential fatty acid, linoleic acid¹. As the magnitude of unsaturation is more in sunflower oil, it is susceptible for rapid autooxidation during storage. Chapmna *et al*². have reported that the fatty acid composition of oil varies with environment. Oils obtained from plants grown under different environmental conditions may show different storage behaviour. Studies have been made on the storage of watermelon seed oil and muskmelon seed oil. Studies have shown that crude groundnut oil stores well in glass, aluminium and tinnedbrass containers. But literature on studies conducted on the storage quality of sunflower oil is scarce under Indian conditions. Yousuf Alikhan *et al*³. have studied the storage of bulk quantities of raw sunflower and groundnut oils. Storage study in small glass containers (as may be useful for household purposes) of crude sunflower oil samples extracted from seeds grown under different climatic conditions was taken up and the results are reported.

Materials and Methods

Ten sunflower oil samples obtained from five different varieties ('EC 68414', 'EC 68415' SuF1, 'Su F2' and 'K 2') grown under irrigated and rainfed conditions were used for this study. The oil was extracted from sundried seeds by Soxhlet fat extraction using solvent ether⁴. A jet of nitrogen gas was passed through the oils to remove any other matter present. Samples of oil were stored in small screw capped colourless bottles 3/4 th height filled. Oil was withdrawn every thirtieth day and the following analyses were carried out. Lipid hydroperoxides which are the first compounds formed during the course of rancidity were measured in terms of the peroxide value⁵ and results expressed as ml of 0.02N sodium thiosulphate. The volatile carbonyl compounds formed were measured by thiobarbituric acid value⁶ at 400 nm and by anisidine value⁷ at 420 nm. The free fatty acids were measured using a rapid colorimetric method⁸. The oxirane compounds were measured by picric acid test⁹.

Re-ults and Discussion

At 180th day, all samples had developed off-flavour which was confirmed by nine untrained judges. The toxic character of rancid oil has been studied¹⁰.

Peroxide value: The initial peroxide values were found to be higher in irrigated samples than in rainfed samples as shown in Table 1. Several studies have indicated that the levels of unsaturated fatty acids and oils in soybeans are strongly influenced by the environment^{2,11,12,13}. There was significant increase in peroxide value between the initial period and 180 days showing the 0_2 uptake by the oil. Although 'peroxides' are possibly not directly responsible for the taste and odour of rancid fats, their concentration as represented by the peroxide value is often useful in assessing the extent to which spoilage has advanced⁵. Among the five different varieties, the variety, 'SuF₁,' had the highest mean peroxide value and the variety ' K_2 ' the lowest with significant difference between them. The greater the magnitude of unsaturation, the greater is the liability of fat to undergo oxidative rancidity⁵. The mean peroxide values were significantly higher or same in all varieties except for 'SuF'₁, under irrigated than under rainfed conditions. Different varieties showed different peroxide values.

Free fatty acid contents: The initial FFA contents were also found to be higher for irrigated samples than for the rainfed samples as shown in Table 2. Increase in FFA during storage could be noticed as reported by Cox and Pearson⁵. As rancidity is usually accompanied by free fatty acid formation, the determination is often

			Irrigated			Rainfed				
Storage period (days)	EC.68414	EC.68415	SuF1	SuF2	K2	EC.68414	EC.68415	SuF1	SuF2	K2
0	3.25	2.1	1.8	1.9	2.2	1.3	0.85	1.15	1.15	0.60
30	3.50	3.0	2.0	2.5	3.0	5.6	3.00	5.50	3.40	2.00
60	4.00	4.0	2.7	3.5	3.8	5.9	3.90	5.60	3.80	2.80
90	4.00	4.4	3.4	4.6	4.2	6.1	6.90	10.00	6.70	3.45
120	6.45	10.3	7.9	7.8	6.7	6.3	6.95	10.05	7.50	4.10
150	7.00	10.4	9.2	9.1	8.5	6.4	7.70	10.30	8.10	4.25
180	7.50	10.6	9.8	9.1	8.7	6.7	8.10	14.80	11.30	5.05
Mean	5.10	6.39	5.26	5.51	5.31	5.47	5.34	8.20	5.99	3.18
*ml of 0.02N sod	lium thiosulg	hate/g of oi	1							

TABLE 1. PEROXIDE VALUE* OF SUNFLOWER OIL COLLECTED FROM CROPS GROWN UNDER IRRIGATED AND RAINFED CONDITIONS

 TABLE 2. FREE FATTY ACID CONTENT (% OLEIC ACID EQUIVALENT) OF SUNFLOWER OIL COLLECTED FROM CROPS GROWN UNDER IRRIGATED

 AND RAINFED CONDITIONS

Storage period (days)		Irrigated					Rainfed				
		EC.68414	EC.68415	SuF1	SuF2	K2	EC.68414	EC.68415	SuF1	SuF2	К2
	0	1.24	2.09	0.96	1.12	1.50	0.71	1.70	0.76	1.02	1.30
	30	1.95	2.65	1.41	1.36	2.16	1.12	2.23	1.05	1.53	2.44
	60	5.49	4.93	1.59	1.53	3.82	2.09	3.71	1.72	2.89	4.47
	90	10.80	10.62	2.75	3.23	7.64	2.41	3.71	1.81	2.89	4.47
	120	10.97	11.00	2.82	3.40	7.64	2.57	3.71	2.00	2.89	4.55
	150	11.68	11.22	2.99	3.51	7.64	2.74	3.82	2.00	2.89	4.55
	180	11.68	11.22	3.35	3.57	7.81	2.74	4.35	2.10	2.97	4.61

TABLE 3. THIOBARBITURIC ACID VALUE (0.D. AT 400 NM) OF SUNFLOWER OIL COLLECTED FROM CROPS GROWN UNDER IRRIGATED AND RAINFED CONDITIONS

Storage period	Irrigated				Rainfed					
(days)	EC.68414	EC.68415	SuF1	SuF2	K2	EC.68414	EC.68415	SuF1	SuF2	K2
0	0.03	0.06	0.11	0.09	0.11	0.02	0.12	0.05	0.03	0.05
30	0.10	0.13	0.15	0.02	0.11	0.6	0.15	0.10	0.09	0.10
60	0.15	0.18	0.21	0.17	0.17	0.18	0.36	0.50	0.27	0.14
90	0.19	0.31	0.23	0.25	0.40	0.20	0.42	0.51	0.44	0.18
120	0.19	0.35	0.35	0.41	0.23	0.46	0.46	0,56	0.58	0.35
150	0.35	0.50	0.36	0.55	0.56	0.29	0.49	0.65	0.61	0.36
180	0.38	0.63	0.36	0.62	0.59	0.34	0.54	0.68	0.65	0.40

used as a general indication of the condition and edible quality of oils⁵. The samples grown underirrigated condition showed significantly higher FFA contents than rainfed samples at the time of the development of offflavour. The different varieties showed different FFA values. It has been reported in cotton seed, that both variety and location significantly influence the moisture and FFA contents. The higher FFA content in the irrigated

samples may be due to the higher moisture content available. Several studies have indicated that in soybeans the lipoxygenase activity appears to be genetically controlled^{2,11,12,13}.

Thiobarbituric acid value (TBA): The volatile carbonyl compounds show up in the TBA value¹⁴. Significant increase in TBA values could be noticed between the initial and final periods of storage indicating the development of off-flavour (Table 3). The TBA values were found to be low in the samples analysed. The main volatile carbonyl compounds resulting from the autoxidation of linoleic acid are hexanal and 2,4-decadienal¹⁵. The concentration which could be very important in the formation of rancid off-flavour are very small even in a highly autooxidized sample¹⁴ both in irrigated and rainfed grown samples.

Anisidine value: The anisidine value signals the presence of carboxyl compounds¹⁴. Upto 90th day, the values were zero except for three samples (Table 4). The mean anisidine value of irrigated samples were found to be higher than rainfed samples. Varietal difference could also be noticed. Except for 'EC-68415', the mean anisidine values of the other varieties were significantly higher or same in irrigated than in rainfed conditions.

Picric acid value: Increase in picric acid value during the storage was observed, which indicated the development of rancidity (Table 5). The values were low as has been indicated by Maza¹⁶. Varietal difference could be noticed. No significant difference could be noticed in oils obtained from irrigated and rainfed crops. The oil samples were grouped into irrigated and rainfed and the discriminant function was worked out with the five different chemical tests. The mean values of the results of the different chemical tests of both irrigated and rainfed samples are given in Table 6. The discriminant function for this is as follows:

$$z=(X_1) + (1.52X_2) - (16.17X_3) + (39.17X_4) - (243.08X_5)$$
 where,

 $X_1X_2X_3, X_4$ and X_5 stand for peroxide value, free fatty acid content, thiobarbituric acid value, anisidine value and picric acid value respectively.

TABLE 4. ANISIDINE VALUE (0.D. AT 420 NM) OF SUNFLOWER OIL COLLECTED FROM CROPS GROWN UNDER IRRIGATED AND RAINFED CONDITIONS

Storage period	Irrigated					Rainfed				
(days)	EC.68414	EC.68415	SuF1	SuF2	K2	EC.68414	EC.68415	SuF1	SuF2	K2
90	0.013	0	0	0	0	0.034	0	0	0	0.035
120	0.038	0	0.076	0.213	0.099	0.056	0.051	0.022	0.099	0.046
150	0.038	0	0.087	0.235	0.099	0.056	0.081	0.058	0.133	0.046
180	0.038	0.024	0.099	0.255	0.099	0.056	0.164	0.096	0.175	0.049
Mean	0.038	0.008	0.087	0.234	0.099	0.056	0.099	0.59	0.136	0.047

Anisidine value was zero in all varieties under irrigated and rainfed conditions upto 60th day of storage.

TABLE 5. PICRIC ACID VALUE (O.D. AT 490 NM) OF SUNFLOWER OIL COLLECTED FROM CROPS GROWN UNDER IRRIGATED AND RAINFED CONDITIONS

C	Irrigated					Rainfed				
(days)	EC.68414	EC.68415	SuF1	SuF2	K2	EC.68414	EC.68415	SuF1	SuF2	K2
60	0	ŋ	0	0	0	0.013	0.027	0.026	0.006	0.016
90	0.029	0.003	0.019	0	0	0.033	0.036	0.042	0.015	0.018
120	0.046	0.039	0.030	0.030	0.029	0.033	0.036	0.042	0.022	0.018
150	0.046	0.040	0.35	0.031	0.05	0.033	0.038	0.042	0.024	0.018
180	0.050	0.046	0.037	0.031	0.038	0.033	0.038	0.042	0.035	0.018

 TABLE 6.
 MEAN VALUES OF DIFFERENT CHEMICAL TESTS OF IRRIGATED

 AND RAINFED SAMPLES
 AND RAINFED SAMPLES

	Irrigated	Rainfed
Peroxide value	5.540	5.637
Free fatty acid content	5.120	2.640
Thiobarbituric acid value	0.280	0.320
Anisidine value	0.093	0.079
Picric acid value	0.031	0.031

The discrimination is found to be highly significant (P=0.01), indicating that the parameters studied were closely associated with the development of rancidty of the oil.

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Organochlorine Pesticide Residues in Groundnut Oil

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Samples of groundnut oil were randomly collected from the markets in Lucknow and Sitapur District of Uttar Pradesh, India. Organochlorine pesticide residues were estimated by gas-liquid chromatography using an electron capture detector (³H). Besides DDT, its metabolites DDD and DDE, aldrin and isomers of HCH were detected in significantly high levels in almost all the samples analyzed. A comparison of DDT levels in groundnut oil with that earlier reported in India suggests, a steady state with regard to DDT contamination of food stuffs.

The presence of persistent organochlorine pesticide (OCP) residues, mainly 2, 2-bis (parachlorophenyl)-1, 1, 1-trichloroethane (DDT) and 1,2,3,4,5,6-hexachlorocyclohexane (HCH) in feed stuffs and body tissues in India, is now well accepted. Main source of these environmental toxicants in body tissues appears to be the contamination of food chain. Organochlorine insecticides are lipophilic in nature and therefore, an assessment of fatty foodstuffs for the contamination of pesticide residues is of relevance. Contamination of various oils with DDT and BHC in India has already been reported^{1,2}. Our earlier studies have also shown the presence of DDT, BHC and Aldrin in mustard, coconut and vegetable oils³. Groundnut is an important oil seed and is consumed as kernels or as oil. It is usually attacked by termites and white-grubs in Punjab and other places, and therefore, aldrin and BHC etc. are recommended for the control of these pests⁴. Present study was designed to assess the pesticide contamination of groundnut oil collected randomly from two districts of Uttar Pradesh.

Materials and Methods

Sixty samples of oil from the local market (40 from Lucknow and adjoining areas and 20 from Sitapur and adjoining areas) were collected randomly, during September to November, 1980, and stored at room temperature till analysed, which is generally within 48 hr.

Chemicals used were of Analar grade unless otherwise indicated and checked against ECD (electron capture detector) contamination. n-Hexane used for extraction was purified and tested for pesticide contamination. Other solvents used were double distilled. Care was also taken to avoid glassware contamination by pestic ides. The method used for extraction of pesticides was that of Mills *et al*⁵. n-Hexane was first treated with 1 ml of distilled water in a clean test tube and placed in a liquid air-methanol bath to remove the traces of acetonitrile, if any. The unfrozen phase, that of hexane, was further treated with conc. H_2SO_4 (1 ml) three times, to remove the fat and the cleaned hexane was collected.

Cleaned samples were then analysed by gas-liquid chromatograph, Varian aerograph series "2400" with 3H⁺ detector, under the following operating conditions⁶.

Carrier gas		Purified nitrogen passing thr- ough silica gel & molecular sieve to remove moisture &
		oxygen respectively.
Gas pressure	• •	65 p.s.i.
Flow rate		40 ml/min
Injector temp.		190°C
Column temp.		180°C
Detector temp.		200°C
Attenuation		x^{16}, x^4
Current		10 ^{−9} mA.
Column		Glass spiral column length 6ft., internal dia 1/8' packed with gas chrome Q (80/100 mesh) coated with 1.5 per cent OV- 17+1.95 per cent OV-210 by weight.
Sample	• •	$1.0 - 5.0 \mu$ l

Standards used were obtained from Poly Science Corporation, Illinois (U.S.A.). Residues detected were further confirmed by thin layer chromatography.

Results and Discussion

Levels of aldrin, HCH, DDT and its metabolites detected in the samples are given in Table 1. A compa-

Pesticides	Sitapur Mean±S.E.	Lucknow Mean \pm S.E.
BHC (HCH)	1.306±0.135	1.340±0.319
Aldrin	0.892±0.255	0.290±0.056
p,p`—DDE	1.330±0.178	0.634±0.136
p,p'—DDD	1.158±0.243	0.367±0.051
p,p'—DDT	0.384±0.077	0.858±0.270
∑DDT	2.960±0.396	1.956±0.370

TABLE 1. ORGANOCHLORINE INSECTICIDE RESIDUES (P.P.M.) IN GROUNDNUT OIL SAMPLES FROM LUCKNOW AND SITAPUR DISTRICTS.

HCH includes \ll, β and γ -isomers (lindane). Σ DDT—Total DDT equivalent.

rison of the present findings has been made with those of earlier ones in Table 2.

Residue level of HCH given in Table 1. comprises of alpha, beta and gamma-isomers, the main stereoisomers of hexachlorocyclohexane (HCH). Although the range of contamination of groundnut oil with HCH is more in the samples collected from Lucknow region (0.199-6.421 ppm) than in Sitapur region (0.296-2.360 ppm), there is not much variation in the mean residue levels (1.340 and 1.300 ppm respectively).

Residues of aldrin, in Sitapur district were about 4 times the levels found in Lucknow. It may be mentioned here that aldrin is closely associated with dieldrin, an epoxidation product of aldrin, but it could not be detected as our analytical procedure in the study involved treatment of extracted samples with concentrated H_2SO_4 which precludes dieldrin⁷. However, since tole-

	TABLE 2 DDT RES	SIDUES IN OILS	
Commodity	Place	No. of samples examined	Residue level (ppm)
Groundnut	Delhi ¹	5	5 - 7.1
,,	Hyderabad ²	10	_
Mustard	Delhi ¹	3	22.1 - 25.7
"	Lucknow ³	25	1.4 - 10.9
Sesame	Delhi ¹	3	10.0 - 12.1
Coconut	Delhi ¹	3	9.3 - 10.6
,,	Lucknow ³	25	0.3 - 1.6
Vegetable	17	25	0.55 - 5.4

Superscripts indicate reference numbers

rance limit fixed by the F.D.A. of the U.S.A. for aldrin/ dieldrin in oil and oil seeds is 0.00 ppm, the exclusion of dieldrin has no impact on the study.

Total DDT equivalent which represents rather complete exposure of crop for DDT was also higher in Sitapur (2.960 p.p.m.) than in Lucknow (1.955 ppm). In Sitapur, the total DDT equivalent was contributed by 1.330 ppm of p,p'-DDE and 1.158 ppm of p,p'-DDD (TDE). Level of p,p'-DDT was only 0.384 ppm. However, in Lucknow 1.95 ppm of total DDT equivalent was contributed by 0.634 ppm of p,p'-DDE and 0.366 ppm of p'p'-DDD. The parent compound p.p'-DDT was 0.857 ppm. Therefore, it may be concluded that either the rate of DDT metabolism in groundnut crop of Sitapur was higher than Lucknow or there was chronic exposure.

The suggested daily intake⁸ of fat in the form of vegetable oils like groundnut, sunflower, etc. is 15g which led to about 45 μ g of total DDT equivalent 19 μ g of HCH and 13 ppm of aldrin, as the daily intake in Sitapur, and about 30 ppm of DDT, 19 ppm of HCH and 4.5 ppm of aldrin in Lucknow. Since pesticidies enter into the body mainly through food chain contamination, edible oils may be taken as one of the main sources of pesticide buildup in the body tissues of humans. The possibility indicated by Thakare *et al*¹. that the oilseeds and oil are the sources of contamination of human fat therefore, seems to be correct.

If we compare the present residue level in the groundnut oil with that of the work done about 8 years ago in Delhi¹ and Hyderabad,² the residue level range of DDT seems to be in a steady state.

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Effect of Methyl Parathion on Body Weight, Water Content and Ionic Changes in the Teleost, *Tilapia Mossambica* (Peters)

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Tilapia mossambica fish were exposed to sublethal concentration (0.09 ppm) of methyl parathion for 48 hr. The Changes in body weight and water content were not significant. The sodium, potassium and calcium ions decreased in all the tissues of methyl parathion exposed fishes. These results suggest that the regulation of osmotic balance is more effective through the operation of salt pump rather than water pump.

The indiscriminate use of pesticides can be considered as one of the factors which alters the environment, causing several imbalances in the ecosystem, especially to the denizens of the aquatic media¹. Hence to face such a changed environment, the animal should certainly undergo changes in its body constitution agreeable to that of the altered environment,^{2,3} leading to changes in the physiology of the organisms. Hence, some preliminary investigations were conducted on the fish, Tilapia mossambica exposed to sublethal concentrations of methyl parathion in static water media. The parameters investigated include the changes in the whole body weight, water content and sodium, potassium and calcium ion contents. Though simple, these parameters are of profound importance in maintaining the homeostatic balance under methyl parathion imposed stress condition.

Materials and Methods

The details of the maintenance, acclimation, feeding and determination of LC_{50} value of methyl parathion to *T. mossambica* were described carlier.^{4,5}. LC_{50} was found to be 0.27 p.p.m. for 48 hr⁵. Hence 0.09 p.p.m. concentration of methyl parath on was selected, since it is a sublethal concentration and the fishes weighing 8.0 ± 2.0 g were sorted into six batches of 10 each and exposed for 48 hr. The troughs containing normal and methyl parathion exposed (MPE) fishes were aerated frequently to prevent hypoxic condition of the medium.

Body weight changes were determined at 12 hr intervals. The water content was determined by drying them in a hot air oven at 80°C for 24 to 48 hr. The process was continued till no changes in weight was observed. The muscle, gill, liver and brain tissues of normal and MPE fishes were wet ashed⁶ and sodium, potassium and calcium were estimated using Elico flame photometer. The results were analysed by student 't', test.⁷

Results and Discussion

The body weight of the MPE fish showed a slight decrease throughout the study up to 48 hr (Table 1), which may be either due to the utilisation of organic reserves or due to the loss of body water or may be due to both. A similar loss of body weight was reported in the house sparrows treated with DDT⁸, fish⁹ and snails¹⁰ under sublethal concentrations of malathion exposure. Since the loss of weight is associated with susceptibility to pesticides¹¹, the gradual decrease in the body weight suggests the possibility of fatality on prolonged exposure, probably due to the loss of some body constituents. Since the water content showed fluctuations (Table 1), the progressive loss of body weight during methyl parathion exposure may involve the loss of some body constituents other than water. The un-

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	12 hr (12 hr exposure		24 hr exposure		36 hr exposure		xposure
Constituent	Normal	MPE	Normal	MPE	Normal	MPE	Normal	MPE
Body weight	10.175	10.095	10,162	10.008	10.152	5.932	10,125	9,757
(g)	± 1.138	±1.129	± 1.038	±1.096	\pm 0.988	±088	±1.023	±1.087
		(0.8)		(-2)		(-2)		(-4)
Water content	0.718	0.706	0.712	0.704	0.714	0.723	0.716	0.728
(g/g fish)	± 0.007	±0.012	±0.009	± 0.008	\pm 0.010	±0.009	±0.009	± 0.012
	_	(-2)		(-1)		(+1)		(+2)

 TABLE 1. CHANGES IN THE BODY WEIGHT AND WATER CONTENT OF NORMAL AND METHYL PARATHION (MPE) EXPOSED FISH FOR

 DIFEFRENT HOURS

Values are mean \pm S.D. of 6 individual observations.

None of them were statistically significant from the control.

Figures in parenthesis indicate decrease (-sign) or increase (+sign) over eontrol.

changed water content in the MPE fishes might also suggest the maintenance of osmotic balance during the stress condition. It is likely that this osmotic balance is the result of salt pump rather than water pump from the body to the medium, aimed towards maintaining the structural and functional properties of the cell under methyl parathion exposure.

The Na⁺, K⁺ and Ca⁺⁺ in the muscle, gill, liver and brain tissues of 48 hr MPE fishes showed a decrease (Table 2). Except for the K⁺ and Ca⁺⁺ contents in muscle and brain tissues, the decrease in Na⁺, K⁺ and Ca⁺⁺ contents in all the four tissues were statistically significant. Eisler and Edmunds¹² reported a decrease in the Na⁺, K⁺ and Ca⁺⁺ contents of the liver of fish exposed to endrin. The decrease in the ionic content of MPE tissues evidently suggest the change in the permeability properties of different biological membrane systems under methyl parathion exposure. It is reported that under stress conditions osmotically active substances like amino acids and metabolites like pyruvate and lactate may be on the increase³, perhaps to compensate the loss of inorganic ions. An increase in the tissue amino acid and pyruvate contents were also observed in the same fish species¹³, thus corrobarating with above observation. In the present study, the methyl parathion stress might have resulted in the hypertonicity of the animal tissues and MPE fish seem to prefer the salt pump over the water pump to maintain the isoosmotic

 Table 2. Changes in sodium, fotassium and calcium contents (mm/g wet wt. tissue) in the muscle, gill, liver and brain tissues

 OF NORMAL AND METHYL PARATHION EXPOSED (MPE) (48 HR) FISH

	Mus	Muscle		Gill		/er	Brain	
Constituent	Normal	MPE	Normal	MPE	Normal	MPE .	Normal	MPE
Sodium	0.0448	0.0380	0.0729	0.0657	0.0410	0.0372	0.0673	0.0611
	± 0.0034	±0.0029**	± 0.0033	±0.0020***	± 0.0023	±0.0026*	± 0.0033	±0.0023**
		(15)		(10)		(9)	_	(10)
Potassium	0.0662	0.0612	0.0367	0.0333	0.048	0.0428	0.0422	0.0410
	± 0.0073	±0.0045 NS	± 0.0028	<u>+</u> .0.0017*	± 0.0061	\pm 0.0051NS	± 0.0024	± 0.0024 NS
		(9)		(9)		(11)		(3)
Calcium	0.0027	0.0024	0.0380	0.0325	0.0045	0.0038	0.0043	0.0038
	± 0.0005	\pm 0.0004NS	± 0.0027	±0.0027**	± 0.0004	$\pm 0.0003^{+}$	± 0.0012	± 0.0012 NS
		(11)		(14)		(15)		(13)

Values are mean \pm S.D. of 6 individual observations.

*Significant at P<0.05; **Significant at P<0.01; NS, not significant; +Significant at P<0.025; ***Significant at P<0.005. Figures in parenthesis indicate % decrease over control. balance and this might have resulted in a decrease in the ionic content of MPE tissues, without change in the water content of the whole animal.

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OXYTETRACYCLINE FERMENTATION: STRAIN SELECTION OF STREPTOMYCES RIMOSUS FOR IMPROVED PRODUCTION OF OXYTETRACYCLINE

Ultraviolet light and X-rays were used to improve yields of oxytetracycline by *Streptomyces rimosus*. A rapid method to isolate improved strain for oxytetracycline production was employed after the preliminary screening of morphological mutants, but it did not prove efficient as the colony which gave a greater zone of inhibition did not always produce a higher antibiotic titre in liquid media. UV-light at a dose level giving about 8% survival resulted in the production of mutants which enhanced antibiotic activity and gave oxytetracycline production 140% better than the original strain of *S. rimosus*, and was superior to other doses of UV-light and X-rays.

Mutations and selection have been used for many years effectively for the isolation of strains of fungi^{1,2} and actinomycetes³ with improved and higher antibiotic production. The technique has been used for the development of new strains with increased production of oxytetracycline^{4,5}, but without much success. The information is limited and there is always a demand for highly improved strains in the competitive field of industrial production. Many mutagenic agents are used for improved strain selection. In our present study, an attempt has been made to develop high yielding strains of *Streptomyces rimosus*, known to produce oxytetracycline, using UV-light and X-rays as mutagens.

A strain of S. rimosus⁶ producing oxytetracycline was maintained on tryptonc-glucose yeast agar slants. Spore suspension was made with sterile distilled water and the viable spore density was adjusted to 5.5×10^7 counts per ml of the suspension. The mutagenic agents employed in this experiment were ultraviolet light (Philips germicidal tube, 15 Watts, 230-250 m μ wave length) and X-rays (RADON type, 110 KV-and 11.2 mA). One ml of the spore suspension taken in a small petri dish was exposed to UV-light at a distance of 16 cm with frequent shaking. In treatment with X-rays, one ml of spore suspension of an improved strain selected after UV-radiation was taken in carrel flask, and was subjected to exposure with X-rays.

The treated spores were plated on the agar medium and the colonies on the plates were flooded with soft nutrient agar seeded with the test organism, *Bacillus cereus*, to measure the potency⁷ of the colonies, which

were morphologically different from the parent and selected in the preliminary screening as morphological mutants. The morphological mutants, selected from different stages of treatment, were finally tested for antibiotic acitivity by a shake flask process in the fermentation medium consisting of 1 per cent corn starch, 1 per cent soybean meal, 0.3 per cent NaNO₃. 0.1 per cent enzyme hydrolyscd casein, 0.02 per cent CaCO₃ and 100 ml distilled water (pH 7.0). Each 250 ml Erlenmeyer flask containing 40 ml medium was incculated with 2.5 per cent (V/V) seed culture of vegetative mycelium and incubated at 28°C for 4 days on a rotary shaker (150 r.p.m.). The antibiotic concentration in fermentation broth was determined by the conventional cup-plate method⁸ using *B*. cereus as the test organism. The mycelial growth was measured by taking the weight of dry cells kept at $80+5^{\circ}$ C in an oven for 24 hr.

Exposure of spores of *S. rimosus* producing oxytetracycline to ultraviolet light for 60 and 90 min. giving about 8 per cent and 0.04 per cent survival respectively, resulted in the production of morphological mutants among the survivors, of which 20 per cent gave higher yields and 66 per cent exhibited activities of oxytetracycline production. An improved strain obtained from UV-radiation was further treated with 25 and 30 krad of X-rays resulting 98.1 and 99.9 per cent killing respectively. But none of the doses of X-rays was effective in producing superior colony among the morphological mutants from the survivors with the exception of low producers (33 per cent at a dose of 25 kilorads and 66 per cent at 30 kilorads).

The zone of inhibition of the test organism as a measure of the potency of the mutants: The relationship between zone-diameter and the antibiotic potency of some of the morphological mutants is shown in Table 1. It was observed that the colony which gave a greater zone of inhibition did not always give a higher antibiotic titre in liquid media. Some colonies which were overgrown by the sensitive organism gave maximum yield of oxytetracycline in the fermentation broth. The colony size of the isolates was more or less the same and their morphology d d not differ markedly from the parent except some pinkish shade, star-like structure, distinct corners with central hole, uneven surface and serrated edges in some of the colonies. Therefore, the zone-diameter in flooding technique cannot be used as an assured means for quick isolation of high yielding mutants of S. rimosus.

Effect of ultraviolet and X-rays on cultural characters and antibiotic potency of S. rimosus: Effect of UV-

Mutants	UV radiation time (min)	X-ray dose (Kr)	Zone- diameter (mm)	Oxytetracycline in broth (µg/ml)
UV-60/1	60	_	20	37
UV-60/2	60	_	30	37
UV-60/3	60	_	32	77
UV-60/4	60		18	103
UV-60/5	60	—	30	28
UV-60/6	60	_	25	43
UV-60/7	60	-	Nil	32
UV-60/8	60	-	24	28
X-25/141	_	25	17	120
X-25/142	_	25	10	105
X-25/143		25	12	120
X-25/144	_	25	7	77
X-25/145	_	25	Nil	120
X-25/146		25	Nil	120
X-30/141	_	30	15	50
X-30/142	_	30	19	120
X-30/143		30	18	110
X-30/144	_	30	28	120
X-30/145	_	30	Nil	38
X-30/146	_	30	13	90

 TABLE 1. RELATIONSHIP BETWEEN ZONE-DIAMETER AND ANTIBIOTIC

 POTENCY AMONG THE MORPHOLOGICAL MUTANTS AFTER TREATMENT

 WITH UV-LIGHT AND X-RAYS

light, and X-ray exposure of S. rimosus on cultural characters and antibiotic production is shown in Tables 2 and 3. On treatment with ultraviolet light for 60 and 90 min, more than hundred colonies from those surviving the treatment were selected as morphological mutants, out of which only 18 isolates were listed in Table 2, showing 4 as high-yielding and 9 as low-yielding types. On further treatment of an improved strain UV-60/14 with X-rays, 12 colonies were selected at random from the morphological mutants and were tested for antibiotic potency and cultural changes in liquid media, but none exhibited better activities (Table 3). Treatment with ultraviolet light for 60 min. produced superior mutants, which gave 80-140 per cent better production of oxytetracycline than the original strain of S. rimosus. There was no relationship between mycelial growth and oxytetracycline production, but it was noted that all the high-yielding mutants showed more rapid and abundant growth than the parent. Although the colonies of the mutants did not differ so far from the parent, in shake flask, the high-yielding mutants showed variability in cultural characters by granular growth and black pigmentation, and low-producers by flaky mycelium and yellow pigment from the parent strain whose mycelium was of globular type and colour of pigment was brown.

 TABLE 2.
 MUTAGENIC EFFECT ON CULTURAL CHARACTERS AND OXYTETRACYCLINE PRODUCTION BY MORPHOLOGICAL MUTANTS FROM A

 PARENT STRAIN OF S. RIMOSUS EXPOSED TO UV-LIGHT

Mutants	UV radiation time (min)	Mycelial growth (mg/ml)	Morphology in liquid media	Pigmentation (broth colour)	Oxytetracycline in broth (μ g/ml)
Parent	60	6.1	globular	brown	50
UV - 60/4	60	7.6	granular	black	103
UV - 60/5	60	6.2	flaky	yellow	28
UV - 63/11	60	7.4	granular	black	90
UV - 60/14	60	8.3	granular	black	120
UV - 60/18	60	5.4	flaky	yellow	27
UV - 60/20	60	4.8	flaky	yellow	32
UV - 60/24	60	6.7	globular	brown	50
UV - 60/25	60	6.5	flaky	yellow	37
UV – 60/28	60	6.9	granular	black	105
UV - 60/36	60	8.1	flaky	yellow	40
UV - 60/39	60	7.7	granular	black	67
UV – 90/1	90	4.7	flaky	yellow	38
UV – 90/2	90	5.5	flaky	yellow	37
UV – 90/4	90	6.9	globular	brown	58
UV – 90/6	90	6.1	flaky	yellow	32
UV – 90/9	90	7.5	globular	brown	57
UV - 90/15	90	6.1	flaky	yellow	32
UV – 90/18	90	6.2	globular	brown	50
- C					

Mutants	X-ray dose (Kr)	Mycelial growth (mg/ml)	Morphology in liquid media	Pigmentation (broth colour)	Oxytetracycline in broth (µg/ml)
UV - 60/14	25	8.3	granular	black	120
X - 25/141	25	8.9	granular	black	120
X - 25/142	25	8.5	granular	black	105
X - 25/143	25	8.3	granular	black	120
X - 25/144	25	6.2	globular	brown	77
X – 25/145	25	7.7	granular	black	120
X - 25/146	25	6.9	granular	black	120
X - 30/141	30	7.9	flaky	yellow	50
X - 30/142	30	8.2	granular	black	120
X - 30/143	30	7.6	granular	black	110
X – 30 /144	30	7.8	granular	black	120
X - 30/145	30	6.4	flaky	yellow	38
X - 30/146	30	7.1	granular	black	90

TABLE 3. MUTAGENIC EFFECT ON CULTURAL CHARACTERS AND ANTIBIOTIC PRODUCTION BY MORPHOLOGICAL MUTANTS FROM A MUTANT STRAIN UV = 60/14 exposed to-x-ray

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PHYSICO-CHEMICAL CHARACTERISTICS AND FATTY ACID COMPOSITION OF SOME IMPORTED AND INDIGENOUS VARIETIES OF RAPESEED OIL

Rapeseed (*Brassica campestris*) oil extracted from eight imported cultivars had erucic acid content ranging from 0 to 3.8 per cent in contrast to 48.8 to 55.4 per cent in indigenous varieties. Distinct differences in Iodine value, saponification value and Bellier turbidity temperature value were observed between imported and indigenous rapeseed oils. Indigenous rapeseed varieties had 7.8 to 13.6 per cent oleic acid, whereas imported varieties had 55.8 to 64.9 per cent. Poly unsaturated fatty acids were present to the extent of 36 per cent in imported oils in contrast to 27 per cent in indigenous rapeseed oil.

Rapcseed (*Brassica campestris*) oil differs from other vegetable oils as they contain 45-55 per cent of erucic acid (cis. 13-dccosencic acid)¹. The use of high erucic acid oil in the diet may cause myocardial fibrosis². New varieties of rapeseed, low in erucic acid, have been developed in Canada and Sweden.

The results of the analysis on the extracted oil from the imported and indigenous rapeseeds are presented in this communication.

Six rapeseed and one oil sample from Canada (var. 'Candle', 'Tower', 'Canola-3', Alberta-2', 'Canola' (*B. campestris* + *B. napus*), 'Saskat chewan', and refined Canola' oil), one refined rapeseed oil from Sweden, one rapeseed (var. 'T-59') and one mustard seed (var. 'RT-11', from Agricultural Research Station, Haryana (India), and one *mustard sample* from Mysore market, were used. The seeds were cleaned by sieving and hand picking. About 50 g seeds in each case was powdered in a waring blender. The oil was completely extracted using petroleum ether (40-60 °C) solvent in a soxhlet apparatus. Solvent was removed under reduced pressure in a rotary evaporator. The filtered oil was used for the analysis. Butyro-refractometer reading (BRR) of the oil samples was determined at 40°C using an oil refractometer (Advance Research Instruments Co., New Delhi). Bellier turbidity temperature (acetic acid method). iodine value (Wij's), saponification value and free fatty acid contents were determined according to AOCS procedures³.

The sodium methoxide catalysed transesterification

TABLE 1. PHYSICAL	AND CHEMICAL C	CHARACTERISTIC	CS OF IMPORTED	AND INDIGENOU	JS RAPESEED OIL	S AND MUSTARD	SEED OIL
Variety	Butyro-refracto- meter reading at 40°C	Refractive index	Bellier turbi- dity temp. (°C)	Iodine value (Wijs')	Saponification value	Unsaponifaible matter (%)	F.F.A (as oleic acid) (%)
Candle	61.0	1.4667	15.5	121.3	187.3	1.50	0.96
Tower	60.0	1.4659	16.0	118.2	191.2	1.42	0.86
Canola-3	60.0	1.4659	17.0	121.2	182.8	1.40	1.54
Alberta-2	60.0	1.4659	16.5	121.5	190.0	1.45	1.40
Canola	60.0	1.4659	17.08	121.4	189. 9	1.40	0.70
Saskatchewan	58.0	1.4646	17.0	115.7	196.0	1.42	0.45
Canola oil (refined)	60.0	1.4659	16.5	121.0	190.0	1.20	0.20
Rapeseed oil (Sweden)	60.0	1.4659	17.0	120.0	189.5	1.30	0.20
T-59 (Haryana)	58.0	1.4646	28.3	106.0	170.2	1.00	0.90
RT-11 (Haryana)*	58.5	1.4636	27.1	106.0	170.5	0.97	0.82
Mysore sample*	59.0	1.4652	28.5	107.0	171.4	0.97	0.98
*Mustard oil sample	es						

TABLE 2. PATTY ACID COMPOSITION OF RAPESEED OIL EXTRACTED FROM IMPORTED AND INDIGENOUS VARIETIES AND MUSTARD SEED OIL

		Fatty acid (%) by wt					Iodine	value					
variety —	Interv Interv 14:0 16:0 16:1 18:0 18:1 18:2 18:3 20:0 20:1 22:0 22					22:1	Experi- mental	Theore- tical					
Candle	Т	3.27	0.35	1.20	55.84	22.83	12.04	0.44	1.35	т	2.50	121.3	121.8
Tower	Т	3.80	0.40	1.21	64.88	16.30	12.07	0.25	1.18	Т	Nil	118.2	117.1
Sample-3	Т	4.05	Т	0.77	53.73	24.56	11.82	0.21	2.48	Т	2.36	121.2	123.6
Alberta-2	Т	3.55	Т	1.01	58.87	21.12	13.53	0.20	1.83	Т	1.89	121.5	123.9
Canola	Т	3.59	0.46	0.81	53.94	24.92	11.95	0.36	2.03	Т	1.84	121.4	124.5
Saskatchewan	Т	4.40	0.35	0.64	59.43	23.03	9.55	0.44	1.46	Т	0,60	115.7	118.1
Rapeseed oil (Sweden)	Т	5.90	Т	1.80	58.90	20.20	13.20	т	т	Т	Nil	121.0	120.3
Canola oil (refined)	Т	5.80	Т	1.30	61.20	14.20	13.20	т	0.50	Т	3.80	120.0	115.5
T-59 (Haryana)	Nil	2.90	Т	1.80	10.10	14.10	10.70	0.50	4.00	0.42	55.40	106.0	106.7
RT-11 (Haryana)*	Nil	2.40	Т	0.70	7.80	15.60	10.20	0.56	4.90	0.46	55.40	106.0	107.0
Mysore samples*	Nil	2.80	т	1.10	13.60	16.50	10.50	0.60	5.70	0.40	48.80	107.0	107.6
T - Traces, ←	- <0.3	5											

*Mustard oil samples.

procedure⁴ was followed for the preparation of fatty acid methyl esters. Separation of the esters was carried out using a CIC dual column chromatograph equipped with hydrogen flame ionisation detector, under the following conditions: Stainless steel column ($5' \times 1/8''$) packed with 15 per cent diethylene glycol succinate on Chromosorb W (80-100 mesh); column temperature 185°C isothermal; injector and detector ports adjusted to 240°C; carrier gas, N₂ 15 ml/min; H₂, 20 ml/min. Identification of the fatty acids was made by comparison with standards (Sigma). Quantitation of the peaks was done by triangulation. Each analysis was carried cut in triplicate and the average is reported.

The results presented in Table 1 show that iodine value of the imported rapeseed oils is higher (115.7 to 121.5) than the oils from indigenous varieties of rapeseed and mustard oils (106 to 107). This is reflected in the fatty acid composition presented in Table 2. Bellier turbidity temperature value ranged between 15.5 and 17.0°C for imported rapeseed oils, whereas it was between 27.1 °C and 28.5 °C for indigenous rapeseed and mustard oils. Unsaponifiable matter was present upto 1.5 per cent in the imported rapeseed oils, in contrast to 1.0 per cent in indigenous rapeseed oils. In general, as can be seen from Table 2, low erucic rapeseed oils contained higher oleic, linoleic and linolenic acids. This is in accordance with earlier observation¹. The experimentally determined iodine values were in agreement with the calculated values (Table 2). The lower Bellier turbidity temperature value (below 17°C) obtained for imported rapeseed oils may be attributed to the presence of lower amounts of arachidic and behenic acids than in indigenous oils. Linolenic acid content in imported rapeseed oils ranged between 12 and 13.5 per cent, whereas it was 10.2 to 10.7 in indigenous oils.

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A SIMPLE DROPPING METHOD FOR DETER-MINING THE TEXTURE IN PARBOILED RICES

Optimally cooked, mild to severely parboiled rice samples after 1 hr of cooking are taken in a rectangular box having flap doors at the bottom. On instantaneously opening the flap doors, the grains fall on a glass plate kept 75 cm below the box. The glass plate is marked with concentric circles of 5 cm radial difference and the grains that occupy different circles are collected separately and weighed quickly. More severe the parboiling, greater was the area to which the cooked rice got distributed. Based on the proportion of grains that occupied the first two circles, the parboiled samples could be graded for their texture. Samples were graded as 'tough', when more than half of the grains got distributed beyond 10 cm radius from the central point of drop.

In recent years, there is a spurt in parboiled rice production in the traditionally raw rice milling areas like Punjab and Andhra Pradesh resulting in flooding of the market with different types of parboiled rices. The existing specifications¹ for parboiled rice do not stipulate any criteria for accepting rice based on its palatability characteristics and consequently the traders face with certain problems in disposing of rice in consuming areas like Kerala. Though certain instrumental tests² ⁵ for determining the texture of cooked rice have been developed and some^{6,7} have been used for assessing the texture of parboiled rice, the need for developing a simple test was felt. Hence, the following dropping method is proposed.

Six month old 'IR 20' paddy was soaked in cold (CS)⁸, warm (WS)⁹ and hot (HS)¹⁰ water and steamed at 0 psig for 10 min. One lot of 'IR 20'—WS was steamed at 0 psig for 20, 30 and 40 min also. WS samples of 'TKM 9' and 'Co 25' were parboiled at 80, 90, 100, 110 and 120°C for 10 min¹¹. One lot of paddy was soaked in water at room temperature for 30 min, steamed at 0 psig for 10 min followed by 5 psig for 20 min and 15 psig for 30 min (designated as PP). All these samples were shade dried, dchulled in a Satake grain testing mill and polished in a McGill miller No. 3 to 6 ± 0.1 per cent degree of milling.

Rice was cooked in excess water to its optimal cooking time¹². As the optimal cooking time could not be

determined in 110°C, 120°C, and PP samples by the customary glass-plate opaque-core method, these samples were cooked for 2 min more than the 40-min steamed samples¹³. After the completion of cooking, the excess cook water was drained and the sample transferred to petridish containing a filter paper inside the lid. The petridish was placed in a water saturated chamber and the samples tested after 1 hr of cooking. The equilibrated moisture content of rice upon soaking at room temperature (EMC-S) was also determined¹⁴.

Device for dropping the cooked rice: Fifteen grams cooked rice kept uniformly distributed in a rectangular sample box (A) of $4 \times 4 \times 4$ cm made of acrylic sheet was momentarily dropped over a glass plate (B) of 60×60 cm placed on the floor by releasing the bottom flaps (C) by turning the release lever (D) which supports the flaps (Fig. 1). Concentric circles of 5 cm radial distance are marked on the glass plate from the central point of



Fig. 1. Schematic diagram of dropping device

A. Sample holder, B. Glass plate, C. Bottom flaps, D. Release

TABLE 1. DISTRIBUTION OF COOKED PARBOILED RICE DROPPED FROM 75 CM HEIGHT (%)

Treatment	l circle	II circle	I + II circle
10 min. steamed	61.4	31.3	92.7
20 ., .,	51.3	28.4	79.7
30 ,, ,,	37.6	21.3	58.9
40 ,, ,	24.7	12.3	37.0
Cold soaked	60.0	25.9	85.9
Warm soaked	55.1	22.8	77.9
Hot soaked	25.3	29.0	54.3
Pressure parboiled	19.3	15.1	34.4

drop. The cooked grains were dropped from a height of 50 and 75 cm on the glass plate and the grains that occupied each circle were quickly collected and weighed separately. The test was repeated thrice.

Wide distribution of samples on the glass plate occurred while dropping from a height of 75 cm and hence the results pertaining to 75 cm height alone were considered. The quantity of grains that remained within 10 cm radius from dropping point was considerably low in case of severely parboiled samples (Table 1) indicating their tough textures. In case of 40-min steamed and **PP** samples, the grains occupying the first circle were less than 25 per cent of the samples taken up for the test, whereas,

TABLE 2. COO	OKED RICE FAL AND THE	LING WITHIN THE	FIRST TWO CIRCLES
Parboiling temp. (°C)	Variety	% grains within the Ist two circles	EMC-S (%, d.b.) of polished rice
Kaw	Co 25	91.7	41.9
	TKM 9	80.4	39.6
80	Co 25	78.3	75.8
	TKM 9	66.0	68.9
90	Co 25	68.9	107.7
	TKM 9	65.2	83.9
100	Co 25	59.1	121.6
	TKM 9	56.5	94.1
110	Co 25	44.4	139.6
	TKM 9	50.1	109.9
120	Co 25	41.3	145.4
	TKM 9	45.7	124.8
PP	Co 25	28.9	167.8
	TKM 9	44.2	136.1
	• …		

PP: Pressure parboiling

they were more than 50 per cenr in 10- and 20-min steamed and CS and WS samples. Except in 40-min steamed 110, 120°C and PP samples, in all other cases the grains that occupied the first two circles formed more than 50 per cent of the samples taken up for the test. Negative correlation ($r = -0.936^{**}$) was observed between the proportion of grains that occupied the first two circles and the EMC-S values (Table 2). The quality of the cooked rices parboiled for a longer time or at temperatures above 100°C was poor¹⁵. This test also indicates a similar pattern in quality. While testing optimally cooked raw samples, appreciable quantity stuck to the sides and the flap doors of the box. Because of the simplicity and rapidity of this test, this method would be useful in grading parboiled rice texture at field level.

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EFFECT OF HEAT TREATMENTS ON STABILITY OF ASCORBIC ACID IN COPPER CONTAMI-NATED MILK OF COWS AND BUFFALOES

Losses in ascorbic acid content during heating and storage of cow and buffalo milk contaminated with copper are reported. Compared to cow milk, the losses were significantly less in buffalow milk. During storage for 48 hr at $5\pm1^{\circ}$ C, ascorbic acid was almost completely destroyed in cow milk contaminated with 0.4 ppm copper; the corresponding losses in buffalow milk were 25-40%.

Ascorbic acid is very sensitive to light, heat and copper. Pasteurization destroys appreciable quantities of ascorbic acid in cow milk, especially when it is contaminated with copper¹⁻³. Earlier work⁴ has shown that in buffalo milk, copper is less detrimental to the stability of ascorbic acid. Losses of ascorbic acid during heating and storage of buffalo milk contaminated with copper are reported here. For the purpose of comparison, data on cow milk are presented.

Milk from cows and buffaloes was collected individually from four animals each, directly in brown glass bottles. The milk was pooled in another brown bottle and divided into two portions. Copper sulphate solution was added to one portion to give it an added copper content of 0.4 ppm. Samples with and without added copper were further divided into four parts and subjected to four different types of heat treatments. These treatments were (i) without heating, the samples were kept at room temperature, (*ii*) heating at 63° C for 30 min in a thermostatic water bath (L.T.L.T. treatment) (iii) heating at 71°C for 15-20 sec (H.T.S.T. treatment), and (iv) heating just to boil, followed by immediately cooling to room temperature (27°C) with chilled water. Ascorbic acid was estimated immediately and after a storage period of 48 hr (in a refrigerator) by the method of Pelletier⁵.

The average ascorbic acid content of fresh cow milk was 17.1 mg/l and that of buffalo milk was 22.8 mg/l. Heating reduced ascorbic acid content in cow and buffalo milk as shown in Table 1. The loss was 10.6, 4.7 and 6.8 per cent during LTLT, HTST and boiling, respectively in cow milk. In buffalo milk, the corresponding values were 7.4, 4.2 and 5. 4 per cent respectively.

	Cow m	ilk	Buffallo milk		
Treatment	Uncontami- nated	Contami- nated*	Uncontami- nated	Contami- nated	
Raw milk	17.1	14.8	22.8	20.5	
L.T.L.T.	15.3	11.3	21.2	18.5	
H.T.S.T.	16.3	13.5	21.9	20,2	
Boiling	15.9	13.7	21.6	19.3	

TABLE 1.	ASCORBIC ACID CONTENT (M/G1) OF COPPER	CONTAMI
	NATED RAW AND HEATED MILK	

*Level of copper contamination 0.4 ppm

Presence of copper influenced the degree of loss. With 0.4 ppm copper, the milk samples lost more than 10 per cent of its ascorbic acid content in about 2 hr. After LTLT, HTST and boiling, the losses in cow milk were 34.0, 21.1 and 20.2 per cent and in buffalo milk 18.9, 11.6 and 15.7 per cent respectively. The destruction of ascorbic acid during heating of buffalo milk was less in the presence of copper.

It is reported that pasteurization of milk by LTLT generally results in about 20 per cent decrease of ascorbic acid, while the loss by HTST method is much less. In the present study, even though the destruction of ascorbic acid was less by HTST than LTLT treatment, this difference was not significant either in cow milk or in buffalo milk which are free from copper contamination. In copper contaminated cow milk however, LTLT treatment caused significantly higher destruction of ascorbic acid than the other two heat treatments. Unlike in cow milk, LTLT treatment—compared to the other two treatments-did not destroy significantly higher quantity of ascorbic acid in buffalo milk.

Ascorbic acid content of stored milk as influenced by different heat treatments is presented in Table 2. Considerable destruction of ascorbic acid occurred during storage of milk depending upon the treatments the samples received. In cow milk samples free from copper contamination, about 25 per cent of the ascorbic acid

Table 2. Ascorbic acid content (/mg/1) of copper contaminated raw and heated milk* stored for 48 hr at $5\pm1^{\circ}c$

	Cow	milk	Buffalo milk		
Treatment	Uncontami- nated	Contami- nated	Uncontami- nated	Contami- nated	
Raw milk	13.2	0.9	21.2	15.1	
L.T.L.T.	11.9	0.7	18.7	14.1	
H.T.S.T.	12.6	1.0	19.7	17.1	
Boiling	12.9	9.3	19.1	16.2	

*Ascorbic acid contents of fresh cow and buffalo milk were 17.1 and 22.8 mg/l respectively.

was destroyed during refrigerated storage for 48 hr, irrespective of the initial heat treatments given to the sample. In copper contaminated milk, the destruction of ascorbic acid was almost complete during this period except in boiled samples. The losses in boiled milk were comparatively less, probably due to the sulphydryl group activated during boiling6-7, which acts as an antoxidant8. The losses of ascorbic acid during storage was noticeably less in buffalo milk compared to cow milk, particularly in samples contaminated with copper to the same degree. Ascorbic acid was almost completely destroyed in copper contaminated cow milk during storage for 48 hr, about 25-40 per cent losses were noticed in buffalo milk. This marked difference in the stability of ascorbic acid could be due to differences in the association of copper with the constituents of milk. It has been reported that complexing of copper alters its catalytic activity⁹. Therefore, changes in the association pattern of copper in milk would affect its influences on ascorbic acid.

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ABSORPTION AND RETENTION OF SULPHUR-DIOXIDE IN RAW MANGO SLICES DURING DRYING AND DEHYDRATION

Raw mango slices after blanching were sulphited by steeping the slices in 1.0, 2.0, 3.0 and 4.0 per cent potasaium metabisulphite (KMS) solution for 15 and 30 min. Absorption and retention of SO_2 increased with increasing concentration of KMS solution and duration of steeping. Longer steeping time (30 min) helped better retention of SO_2 .

The presence of sulphur dioxide (SO_2) in the dried fruit is essential to preserve natural colour, flavour, and palatability and to prevent nutritional losses, like ascorbic acid, during drying and storage. It is necessary to standardize conditions, whereby enough SO₂ is taken up by the fruit for proper retention after drying. Sulphur dioxide treatment is done either by burning raw sulphur in the presence of fruits or by dipping the fruits in sulphite, bisulphite, or metabisulphite solutions. The absorption and type of fruit^{1,2}. In case of pineapple, Bhatia *et al*³. observed that steeping in 0.25 to 2.5 per cent potassium metabisulphite solution for 30 min was insufficient to give the desired

concentration of SO₂. McBean *et al*². observed that SO₂ concentration in the atmosphere surrounding the fruit was one of the most important factors influencing the uptake of SO₂ and the absorption of SO₂ increased with increase in time.

To assess the possibilities of utilizing dried raw mango slices for preparation of mango chutney, absorption and retention of SO_2 during drying and dehydration of raw mango slices have been studied.

Raw mangoes of required maturity, size and shape were collected from market, peeled, cut into slices, blanched in boiling water for 2.5 min. for peroxidase inactivation⁴, and then sulphited. Sulphitation was done by steeping the slices in 1.0, 2.0, 3.0 and 4.0 per cent KMS solution for 15 and 30 min. The ratio of weight of potassium metabisulphite solution to slices was 2:1. Sulphited slices were dried in sun and in cabinet drier at $57^{\circ}\pm5^{\circ}C$ temperature to 6 per cent moisture level. Absorption of SO₂ during sulphitation and its retention during drying was determined by the method of A.O.-A.C.⁵.

The amount of the SO_2 absorbed and retained was directly proportional to the concentration of KMS solution and the time of steeping (Fig. 1 and Table 1).

Method of drying	KMS concn (%)	Absorption of SO periods	(ppm) at indicated of steeping	Retention of SO ₂ (ppm) at indicated periods of steeping		
		15 min	30 min	15 min	30 min	
Sun-drying	1.0	1000 (8333.30)	1248 ((10400)	8800 (93.61)	256 (269.47)	
	2.0	1231 (10258.33)	2112 (17600)	500 (544.68)	950 (1000.00)	
	3.0	2258 (18866.66)	3480 (29000)	1504 (1600.00)	2288 (24.08.42)	
	4.0	3027 (25225.00)	4664 (38866.66)	2421 (2575.53)	5472 (5760.00)	
Cabinet drying	1.0	1000 (8333.30)	1248 (10400)	288.00 (303.15)	1056.00 (1111.57)	
	2.00	1231 (10258.33)	2112 (17600)	1440 (1515.78)	2816 (2964,21)	
	3.0	2258 (18866.66)	3480 (29000)	2764 (2909.47)	5488 (5776.84)	
	4.0	3027 (25225.00)	4664 (38866.66)	2816 (2964.21)	5500 (5789,47)	

TABLE 1. ABSORPTION AND RETENTION OF SULPHURDIOXIDE BY RAW MANGO SLICES AS AFFECTED BY CONCENTRATION OF KMS AND TIME OF STEEPING

Figures in parentheses indicate values on moisture free basis.



FIG. 1. Relative absorption and retention of Sulphur dioxide in mango slices.

Similar observations were made by McBcan *et al*². in apricot, peaches and pears, Long *et al*¹. and Kikon⁶ in apples and Hirway⁷ in guava. However, in cabinet drying, above 3 per cent concentration of KMS a corresponding increase in concentration of KMS did not help to have proportional retention of SO₂. Absorption and retention of SO₂ in all the concentration of KMS was greater in slices steeped for 30 min than for 15 min. Between the two methods of drying, the retention of SO₂ was more in cabinet dried slices as compared to sun-dried samples, irrespective of steeping time and concentration of KMS (Fig. 1 and Table 1).

In the present study, the maximum retention of SO_2 was in those treated with 4.0 per cent KMS for 30 min.

This retention was above the upper limit prescribed by F.P.O.8. Therefore, to get the maximum retention of SO_2 of 2000 ppm in raw mango slices, sulphitation should be done for 30 min in less than 3 per cent KMS concentration when it is sun-dried, and less than 2 per cent KMS for cabinet drying. However, with 15 min steeping the concentration of KMS should be increased further.

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BOOK REVIEWS

Developments in Food Preservation 1: Edited by Stuart Thorne., All ed Science Publishers Limited, Ripple Road, Barking Essex, United Kingdom, 1981: pp. 272 Price: £ 30.00.

The book comprises of eight chapters on different aspects on the scope and applications of various methods of food preservation.

First chapter deals with the appropriate technology in food preservation for the developing countries. Reasons for very slow progress of food processing industries in the third world countries have been brought out with concrete suggestions for its improvements like, orienting food processing towards national, rather than western food technology; giving more emphasis on drying and pickling; use of solar energy and bio-mass, as energy sources and utilizing plastic pouches as containers.

Second chapter is on cooling aspects of the horticultural produce. Interactions between heat generation during respiration and cooling effect of transpiration have been discussed. Various factors affecting the cooling process, viz., physiological stage of the product. skin thickness, porosity of packing material, vents in the package, air velocity, air temperature etc. are discussed. More dependence on adiabatic temperature rise, rather than carbon dioxide measurement, while calculating heat generation by hortciultural produce has been emphasised.

Latest methods employed for fruit juice concentration are dealt in the third chapter. Due emphasis has been given to exploit the 'Reverse-osmosis' process for production of semi-concentrates from fruit juices due to its inherent advantages. Due to loss of low molecular weight, water soluble aroma components during concentration decrease by this process; author has cautioned to restrict its use to fruits with high acidity and less aroma.

The advantages as well as limitations of microwave processing of foods is dealt at length in chapter four. Usefulness of this heating technique for enzyme inactivation, vegetable dehydration, fruit juice pasteurization and defrosting of frozen products along with its limitations like uneven distribution of the energy inside the cooker and non adjustable nature of the cooker to meet the energy requirements of individual produce have been discussed. Author is of the opinion that though at present, complexity of many foods and biological variations do not assist the application of microwaves, standardization of the cookers followed by the stand-

dardization of cooking procedures in a scientific manner would enhance its consumer acceptance.

Chapter five deals with the principle and practices of freeze drying of foods, the freeze drying equipments as well as the economic aspects of this method of preservation. Merits of freeze dried products and need for the proper packaging and storage of such products have been emphasised.

Chapter six deals with 'Extrusion processing'. This most complex system has been presented in a very simple way for proper understanding. The biopolymer aggregate changes are discussed and effect of common ingredients like salt and sugar has been explained using heuristic and hypothetical thearies. Different types of extruders used by the food industry are discussed and the difference in their functional performances are brought out.

The editor himself is the author of the seventh chapter dealing with the effect of temperature on the deterioration of stored agricultural produce. The reactions and the processes responsible for the quality changes during storage of fruits, vegetables and other farm produce have been brought out well in this chapter. The most valuable information furnished by the author is the effect of fluctuating temperatures on the storage changes and the reasons for the differential behaviour of different commodities during storage.

The last chapter is devoted to thermal sterilization of foods. Microbial and enzymatic responses to this method of preservation are explained giving specific examples. Author has brought out the advantage of using stork continuous sterilizer (named as Hydromatic sterilizer) for thermal sterilization of foods. Drastic reduction in the consumption of both steam and water due to deployment of regenearion process is the main advantage of this process, and steam saving upto 60-70 per cent is possible in this case.

This book is very informative, comprehensive and upto-date. Its get up is very good with 37 tables and 99 illustrations evenly distributed and properly presented making the text interesting. The brief summaries of the chapters give adequate idea regarding contents and utility of the text to the reader. A good biblicgraphy comprising of 358 references on the subjects of vital importance in the field of food preservation will be very useful to not only food scientists and technologists, but also to the food processors.

K. L. CHADHA Indian Institute of Horticuliural Research., Bangalore. Food Industry Wastes: Disposal and Recovery: Edited by A. Herzka and R. G. Booth. (Ed) Applied Science Publishers, London & New Jersey, 1981, pp. 246; Prize £ 20.00.

This book is the outcome of a symposium organised by the Association of Consulting Scientists, held in Norwich, U.K. from 11 to 13 Nov. 1980. It contains 17 papers on different aspects of waste disposal and recovery from various food Industries. The first threequarters of the book deals with methods for separation and disposal of waste particularly solids frcm foodbearing effluents and the destruction of the solubles in suspension or solution by aerobic or anerobic methods. The latter part deals with processes for recovery of useful food or feed materials from effluents and the feasibility of such processes financially. In the first three papers of the book, the situation relating to legislation and charges for disposal to sewers are considered.

Useful practical details are given for the treatment of specific wastes from dairies, breweries and distilleries, meat and animal by-products, by processes ranging from dissolved air floatation to protein recovery and aerobic treatment of the final irreducible minimum of wastes.

Problems connected with high BOD (Biological Oxygen Demand) food effluents have received detailed attention in recent years, mainly due to increasing awareness of pollution effects, energy price trends and the potential of resource recovery. The two major modern methods of recovery of souble proteins, namely. ultrafiltration and adsorption have been described with sufficient details of membranes and membrane support systems in the case of ultrafiltration and the various adsorbants and methods of using them with concomittant advantages and disadvantages in the latter case. The two methods have been regarded as complimentary rather than competitive. One paper on, recovery of fruit and vegetable waste, describes the possibility of converting the waste into high grade food-stuff biomass with 20-30 per cent protein or converting it into a fuel suitable for many uses.

The authors of the papers are mostly consultants with specialised knowledge in different food industries and have made realistic assessments of the problems associated with effluents. This is a very useful book on a subject of growing importance.

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Developments in Food Carbohydrate 3.: Edited by C. K. Lee and M. G. Lindley, Applied Science Publishers, London, 1982; pp. Xii+217. Price, £ 26.00.

This book records a comprehensive review on disaccharidases, the carbohydrate degrading enzymes. The articles, clearly written by several outstanding authors, are grouped into eight individual chapters with an index at the end.

The book begins with an article covering some of the recent advances on the biochemical nature, properties/ function, and applications of invertase from a variety of sources. The discussion on protein conformational stability of the enzyme is beautifully dealt with. Following this, is a review of lactase, a commercially important enzyme in the dairy and allied industries. Lactase plays a major role in lactose intolerance and methods to alleviate the same are ably presented. The article appears interesting with the inclusion of industrial process schemes for the hydrolysis of milk and whey syrups, as well as several photographs. In the next chapter is presented a discussion on enzymes capable of hydrolysing maltose. Such enzymes, having a very wide substracte specificities, are ubiquitous in nature. Industrially they are of great commercial value in the production of glucose syrups from starch. Trehalase(s), an enzyme of bic-chemical interest is reviewed in chapter 4 in considerable length from an academic point of view. The physiological role of this enzyme is much debated. The succeeding two chapters deal with the biochemistry of several disaccharidases, including their physiological and nutritional significance. The last two chapters describe the various biochemical parameters of several disaccharides of food origin. In addition to functioning as possible energy sources, these disaccharides are very much inovlved in a variety of clinical symptoms manifest in humans (and animals).

In short, the book summarizes a timely and an up-todate account of disaccharidases, which have not been covered so far in a single book of this nature. It is impeccably edited, nicely indexed and bcautifully produced. It is generally free of typographical errors. The generous list of references, as supplements to each article, should make the book a useful reference volume to food scientists and technologists as well as biochemists and nutritionists. The bcok is a valuable addition to the library.

> R. N. THARANATHAN C.F.T.R.I., Mysore.



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- Manuscripts of papers (in triplicate) should be typewritten in double space on one side of bond paper. The manuscripts should be complete and in final form, since only minor corrections are allowed at the proof stage. The paper submitted should not have been published or data communicated for publication anywhere else. Review papers are also accepted.
- 2. Short communications in the nature of Research Notes should clearly indicate the scope of the investigation and the salient features of the results.
- 3. Names of chemical compounds and not their formulae should be used in the text. Superscripts and subscripts should be legibly and carefully placed. Foot notes especially for text should be avoided as far as possible.
- 4 Abstract: The abstract should indicate the principal findings of the paper. It should be about 200 words. It should be in such a form that abstracting periodicals can readily use it.
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- (c) References to article in a book: Joshi, S. V., in the Chemistry of Synthetic Dyes, by Venkataraman, K., Academic Press, Inc., New York, 1952, Vol. II, 966.
- (d) Proceedings, Conferences and Symposia Papers: Nambudiri, E. S. and Lewis, Y. S., Cocoa in confectionery, Proceedings of the Symposium on the Status and Prospects of the Confectionery Industry in India, Mysore, May 1979, 27.
- (e) Thesis: Sathyanarayan, Y., Phytosociological Studies on the Calcicolous Plants of Bombay, 1953, Ph.D. Thesis, Bombay University.
- (f) Unpublished Work: Rao, G., unpublished, Central Food Technological Research Institute Mysore, India.

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